



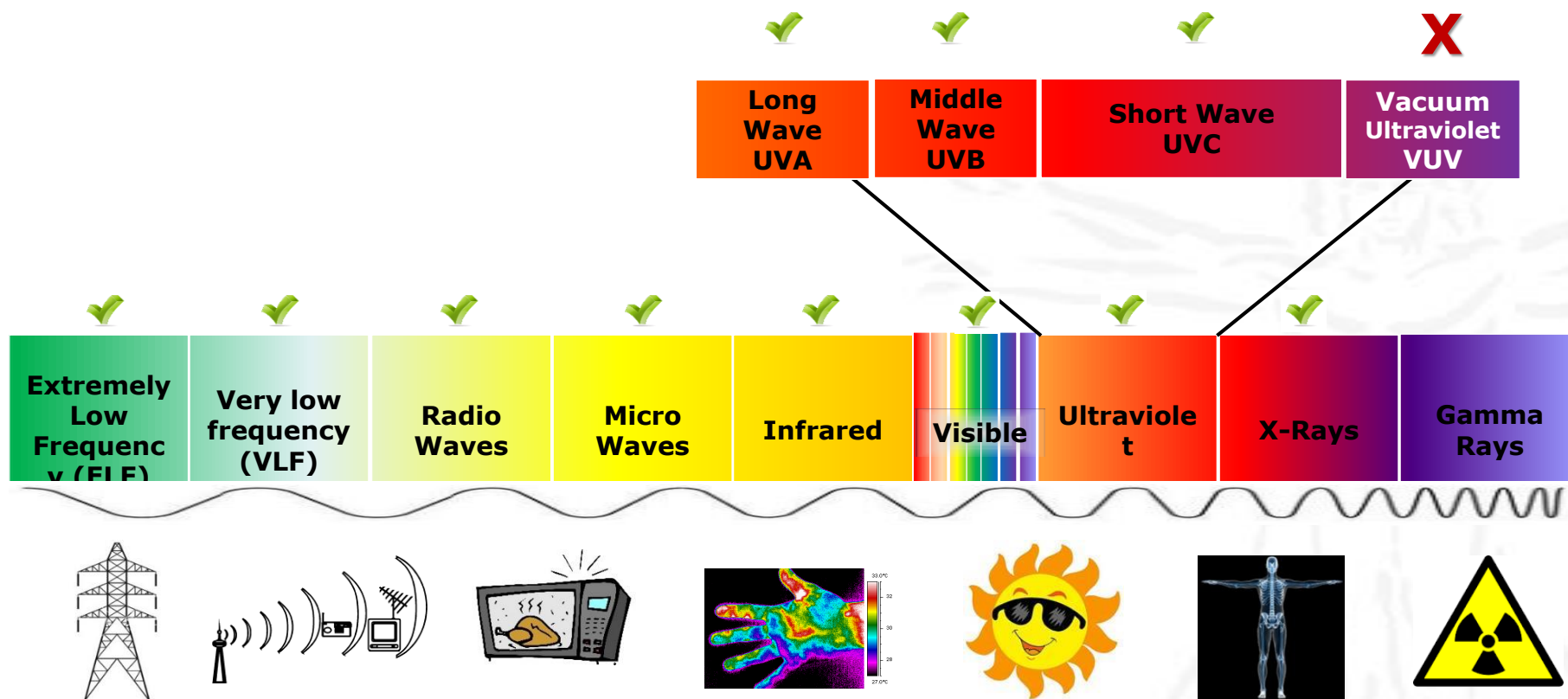
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Vacuum-Ultraviolet Detection

a Completely new Detection Technique for GC

Dr. Sjaak de Koning
Product Manager Advanced Detections
Da Vinci Laboratory Solution

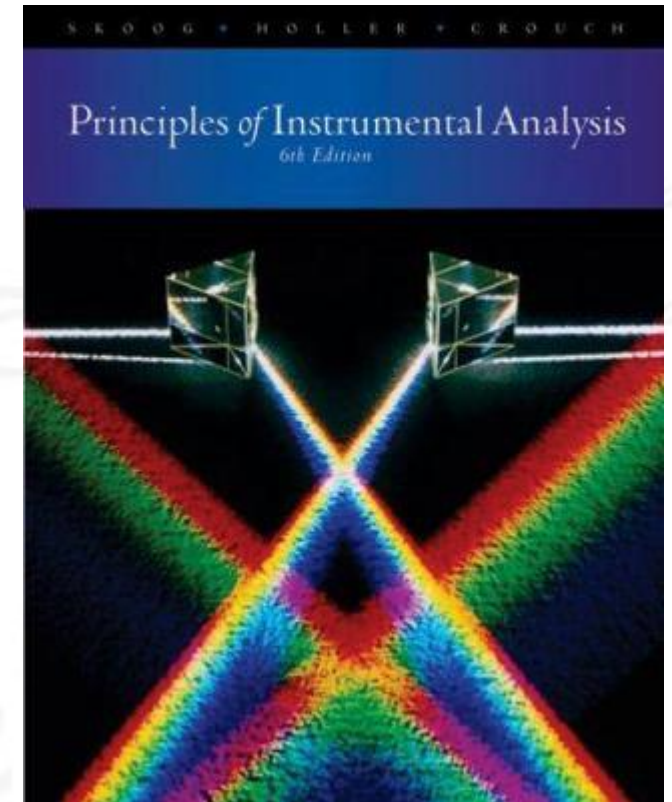
The Electromagnetic Spectrum



- The Vacuum Ultraviolet (VUV) portion of the ultraviolet spectrum is a region of short wavelength, high energy absorbance that provides unique spectral fingerprints of nearly every compound
- He, Ar, H₂ are the exceptions
- Photons in this region can induce electronic transitions, especially ground state to excited state transitions such as $\sigma \rightarrow \sigma^*$ and $\pi \rightarrow \pi^*$
- Compound identification is unambiguous due to the unique nature of the absorbance

Teached at school

- Principals of Instrumental Analysis, by Douglass Skoog, Sixth Edition, 2006
 - "The excitation **energies associated with electrons forming most single bonds are sufficiently high that absorption by them is restricted to the** so-called **vacuum ultraviolet** region ($\lambda < 185\text{nm}$), where components in the atmosphere also absorb strongly. **The experimental difficulties** associated with the vacuum ultraviolet **are significant; as a result** **no further discussion will be devoted to this** type of absorption."

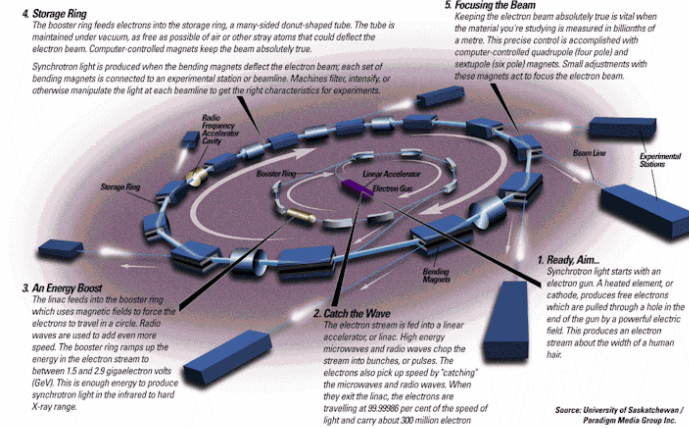


It has been taught to be instrumentally too difficult.

Only in synchotron



How a Synchrotron Works



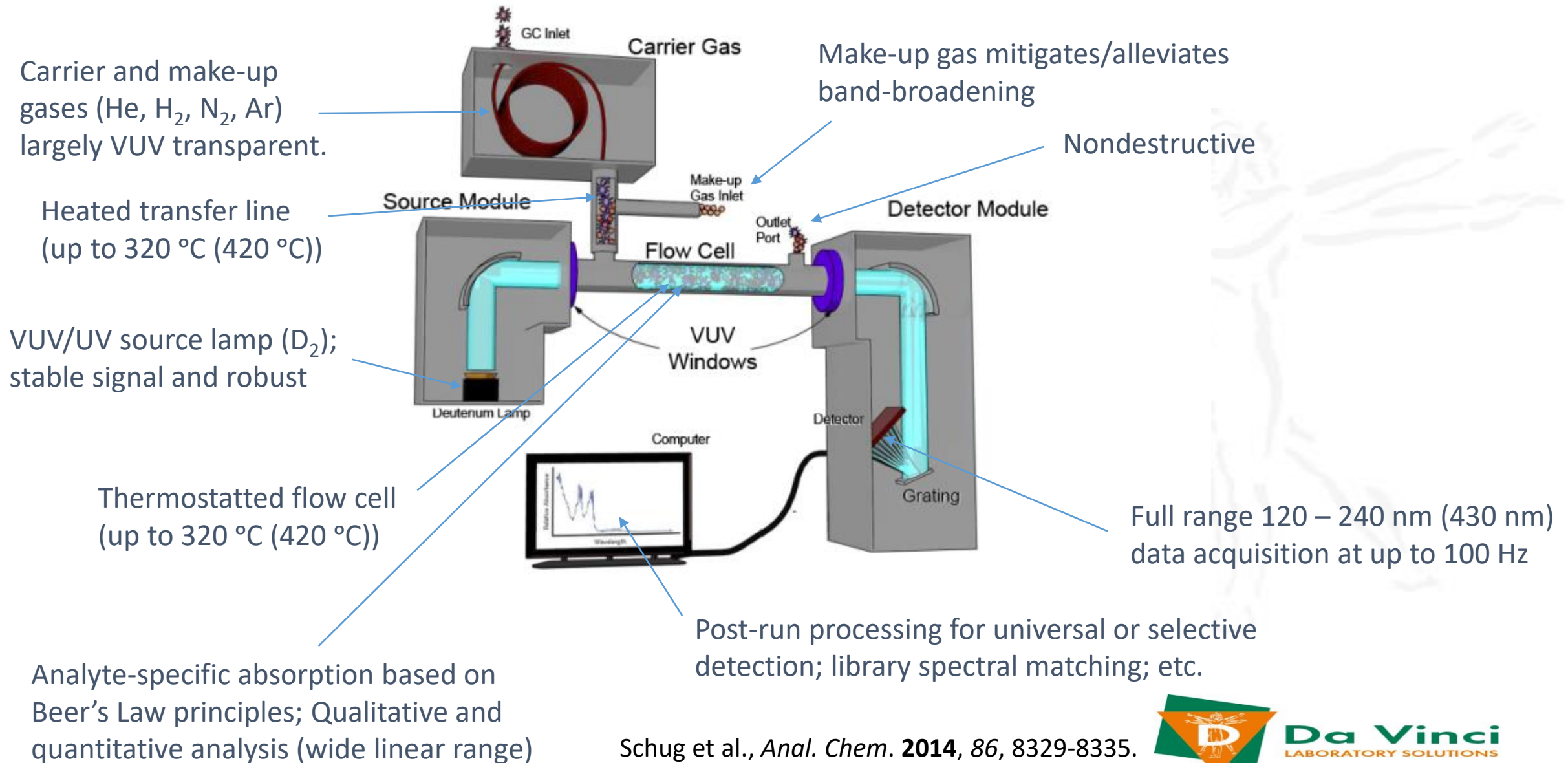
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World's First GC–VUV Detector

- Vacuum Ultraviolet (VUV) Absorption from 120nm to 240nm
 - VGA-101 extends this capability out to 430 nm
- Enables powerful detection capabilities
 - Unique selectivity
 - Unambiguous compound identification
 - Easily deconvolve co-eluting analytes
 - Clear and easy isomer differentiation
 - Excellent sensitivity
 - Low picogram
 - Non-destructive analysis
 - No ionization required
 - Excellent temporal resolution
 - Up to 100Hz Sampling
 - Predictable linear response
 - 1st principle detection reduces calibration burdens
 - Reliable & Easy to use
 - No vacuums pumps

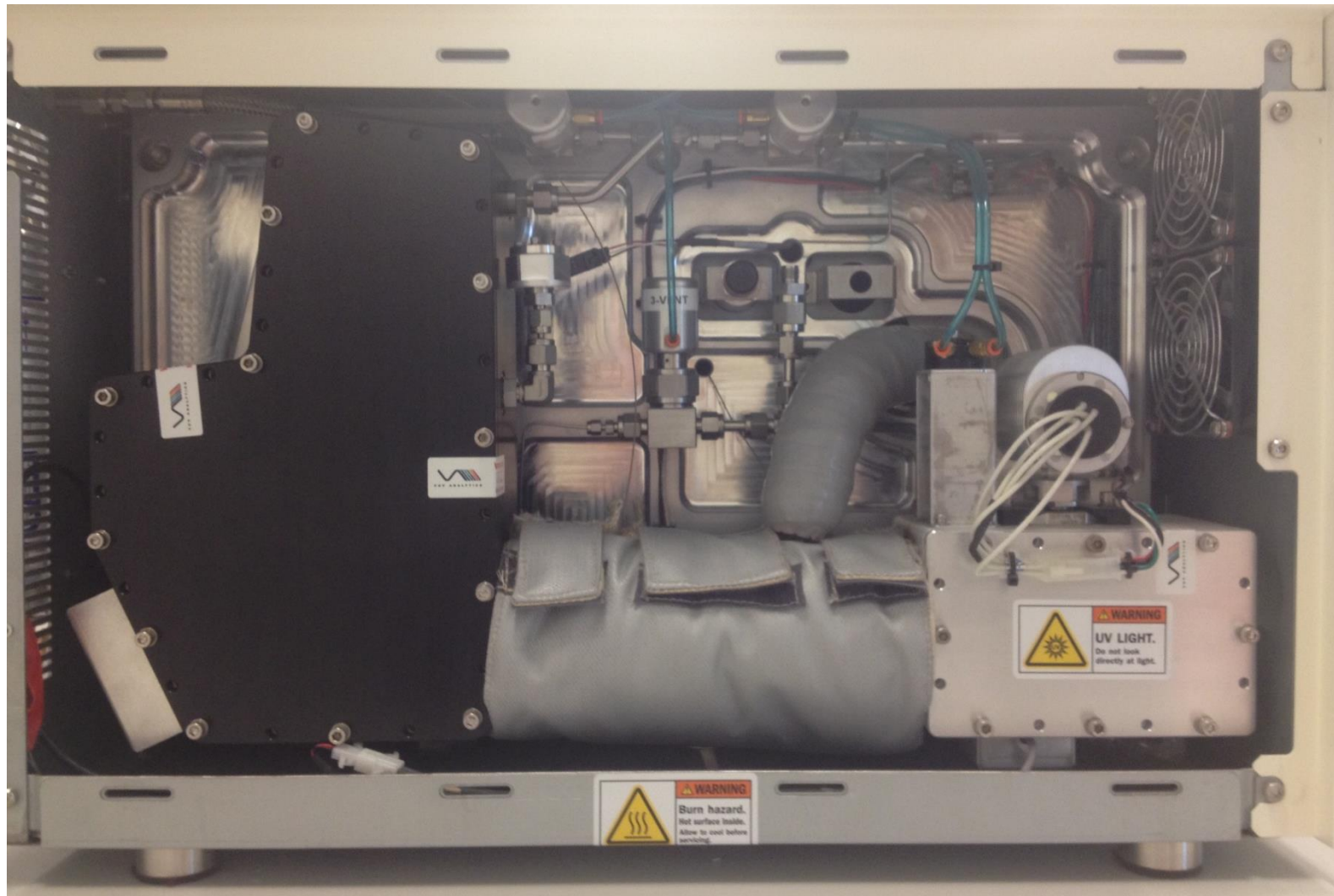


How does it work



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How does it work



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Easy Spectral Quantitation

Governed by Beer's Law:

$$A = \epsilon bC$$

- A = absorbance
- ϵ = extinction coefficient ($M^{-1}.cm^{-1}$)
- b = path length (cm)
- C = molar concentration ($mol.dm^{-3}$)

Peak area of an analyte depends on:

- Concentration of analyte (the number of molecules or mass)
- Cross-section (ability of a molecule to absorb a photon)
- Total flow rate through the flow cell (residence time)
- Flow cell geometry (length and volume)

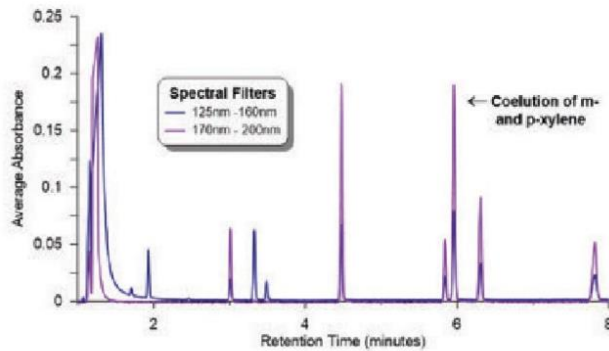
Spectral quantitation is robust and repeatable

- Identity not dependent on retention time or baseline resolution
- Compound % Mass is directly proportional to sum of its absorbance
- No special integration tools required
 - All fitting and quantitation is performed by the software
- VUV absorbance is an inherent property unique to every compound

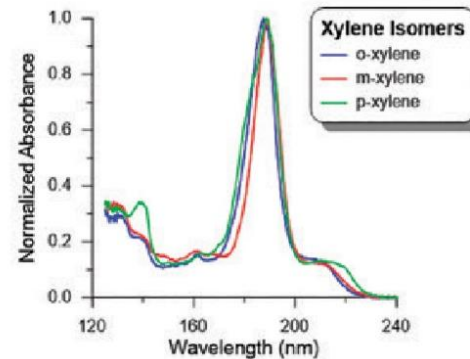
GC–VUV Data is 3D

GC-VUV absorbance data contains both chromatographic & spectral information
(VUV absorbance, time, and wavelength)

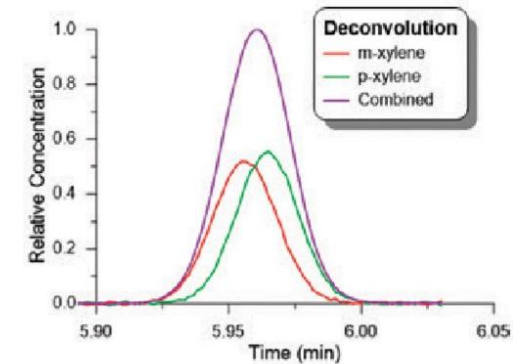
Spectral deconvolution of m&p Xylene



GC-VUV chromatogram showing the co-elution of m- and p-xylene



VUV absorbance spectra of overlapping xylene isomers



Deconvolution of the overlapping chromatographic signals for m- and p-xylene

Chromatographic baseline resolution is not necessary
Co-eluting analytes and isomers can be spectrally deconvolved and quantitated

Some applications

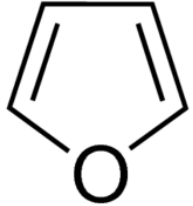


Factory exhaust

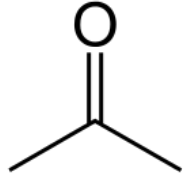


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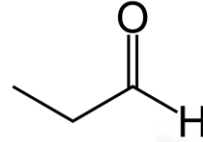
Compounds of interest



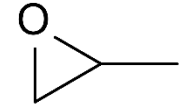
Furan



Acetone



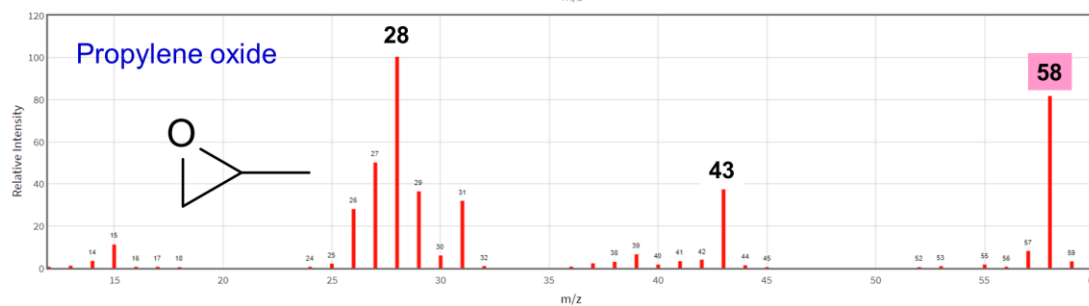
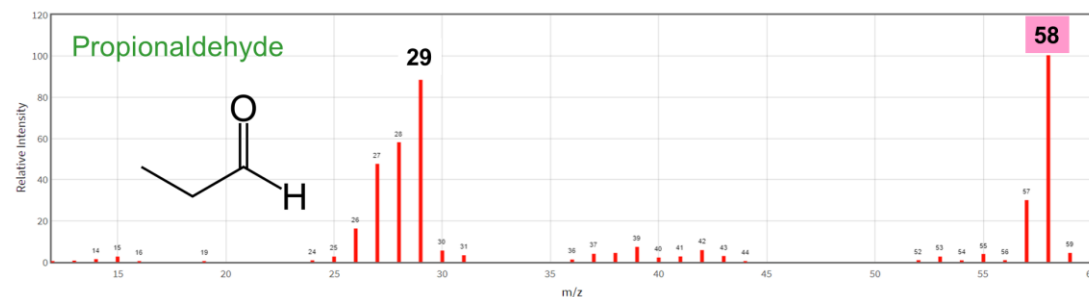
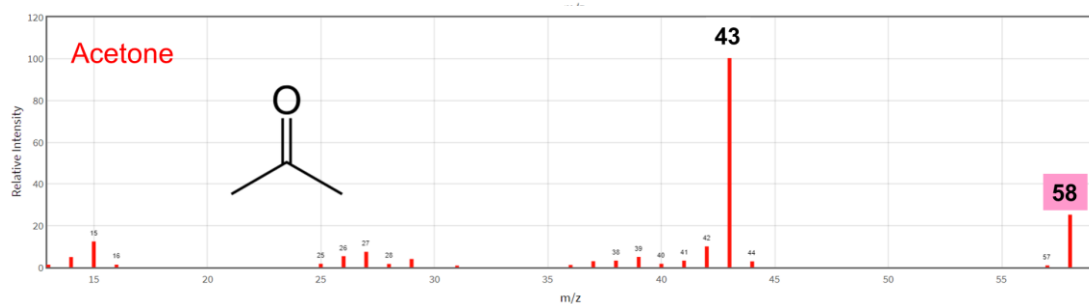
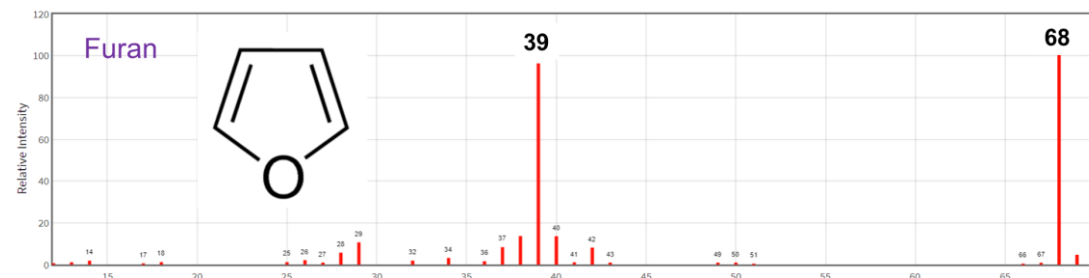
Propionaldehyde



Propylene oxide

- Small molecules
- Coelute under many GC conditions
- Mostly cannot be differentiated by electron ionization MS
- Can they be analyzed successfully using GC-VUV?

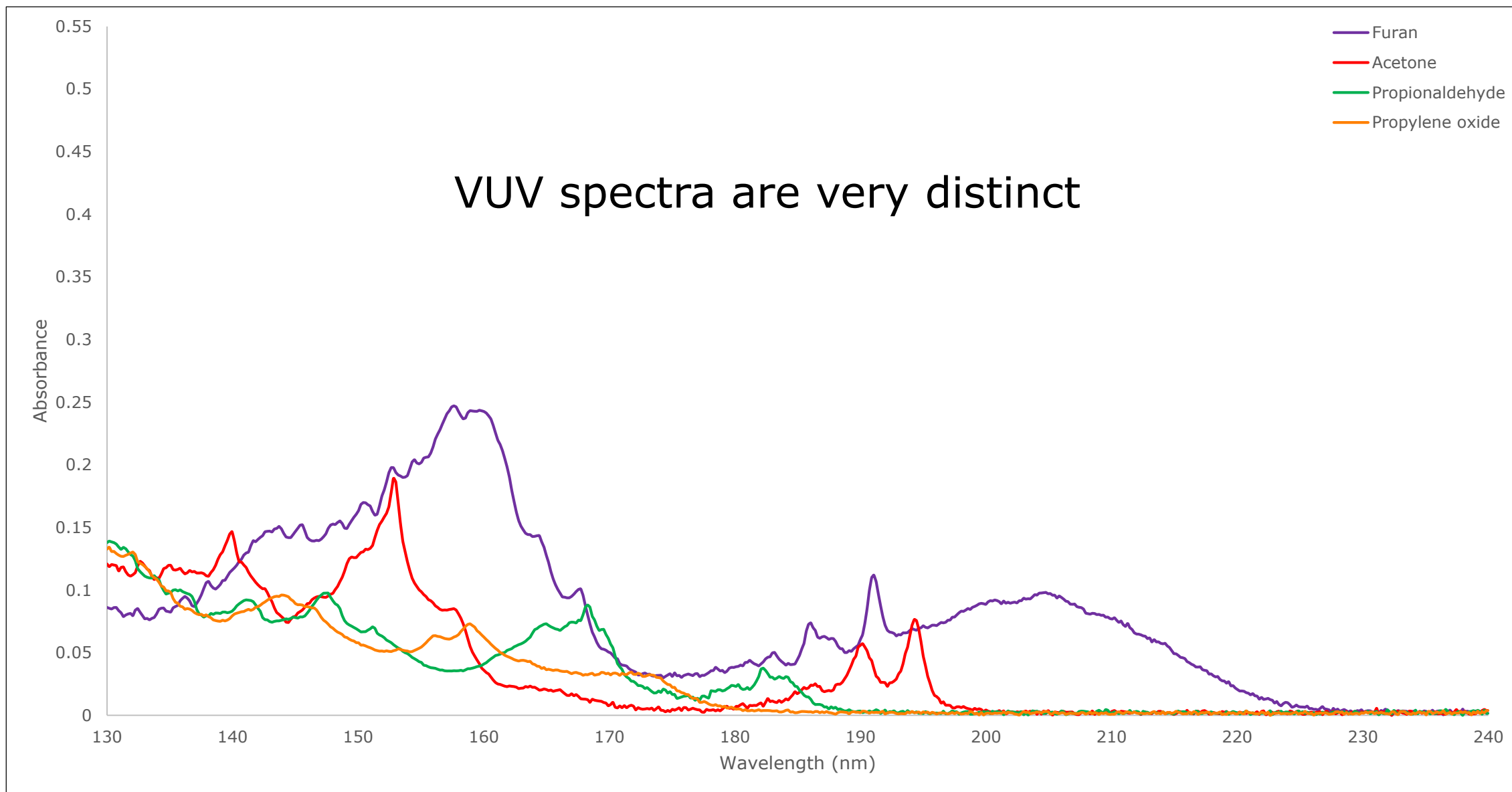
MS spectra



MS spectra are closely related

VUV spectra

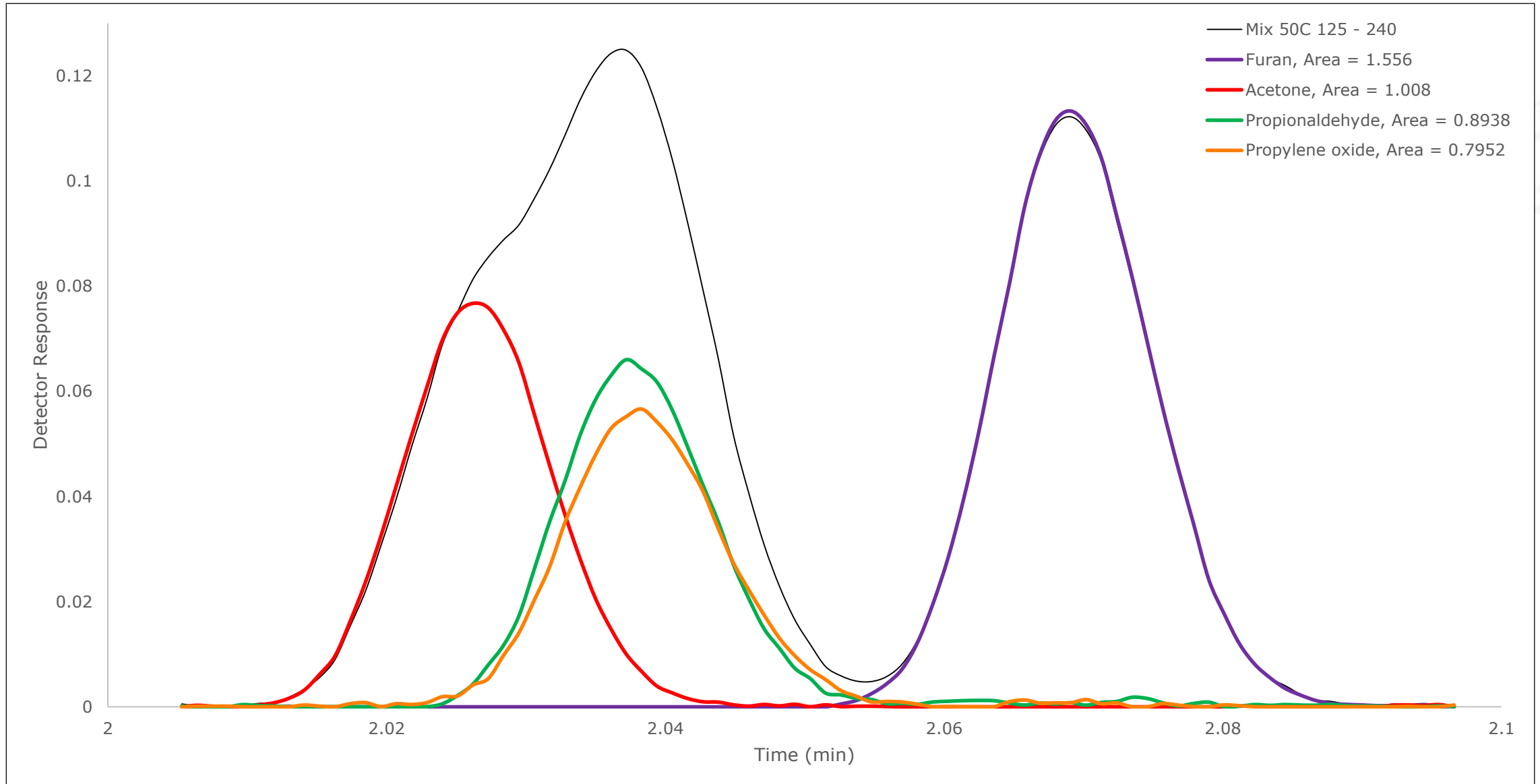
VUV spectra are very distinct



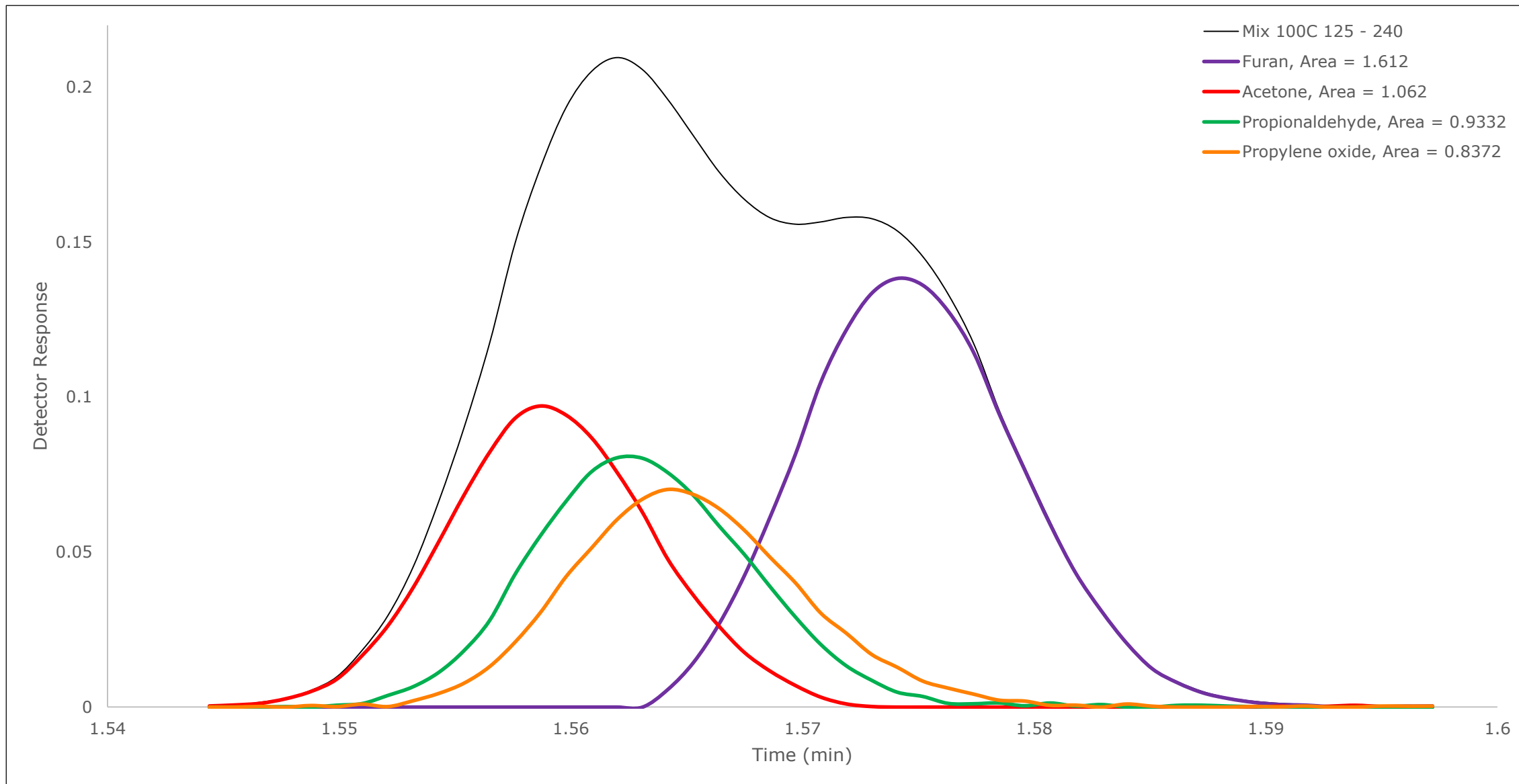
GC—VUV Conditions

- Agilent 6890 GC
 - 30m x 0.25mm x 0.50 μ m Rxi-5ms
 - Column Flow: 2 mL/min He
 - Oven: Isothermal (various temperatures), hold 3 min
 - Inlet: 200° C
 - Split Ratio: 495:1
 - Injection Volume: 0.2 μ L
- VUV Analytics VGA-100
 - Transfer Line and Flow Cell Temperature: 275° C
 - Makeup Gas: 2.83 psi N₂
 - Acquisition Rate: 15 spectra/sec

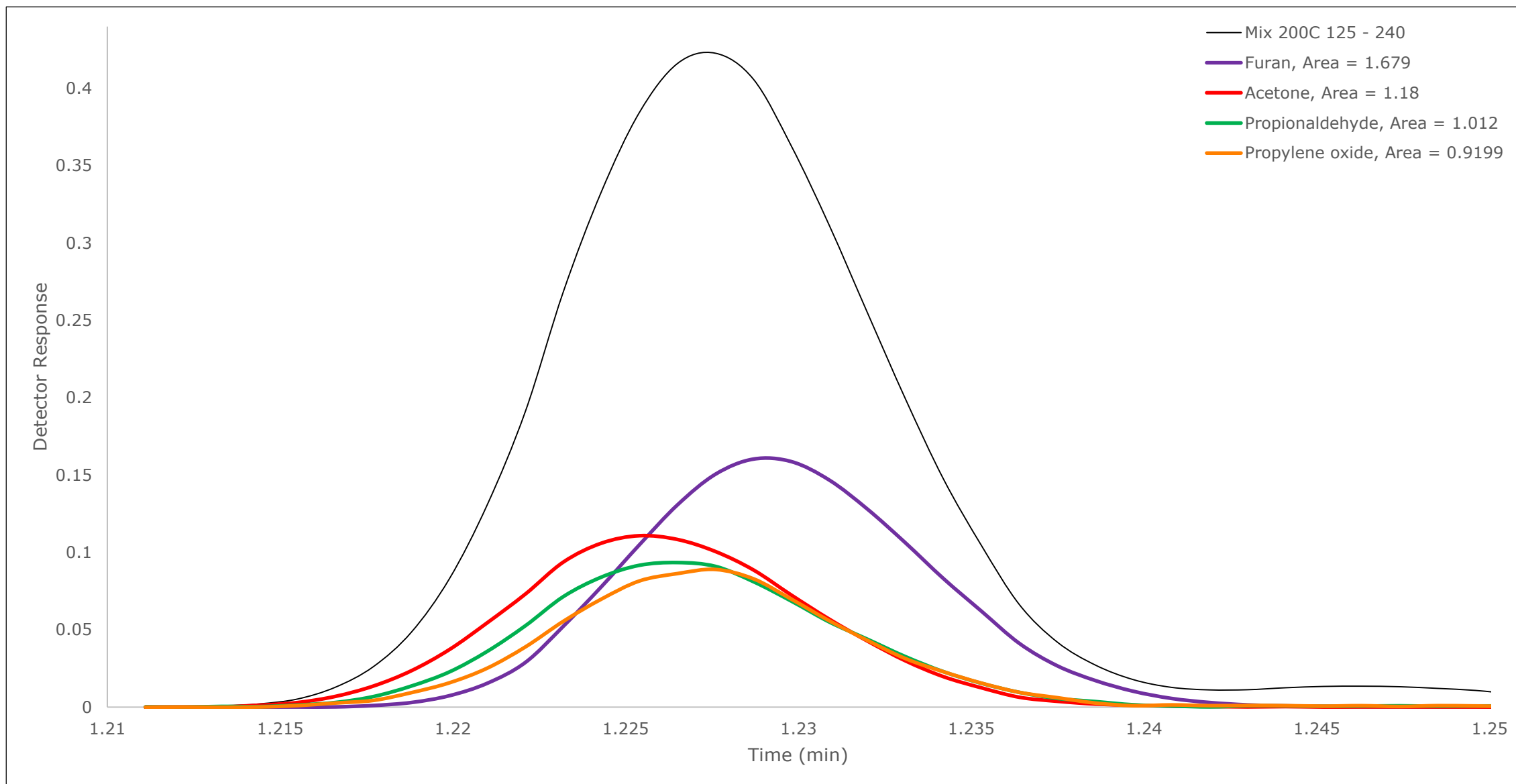
GC Oven Isothermal 50° C



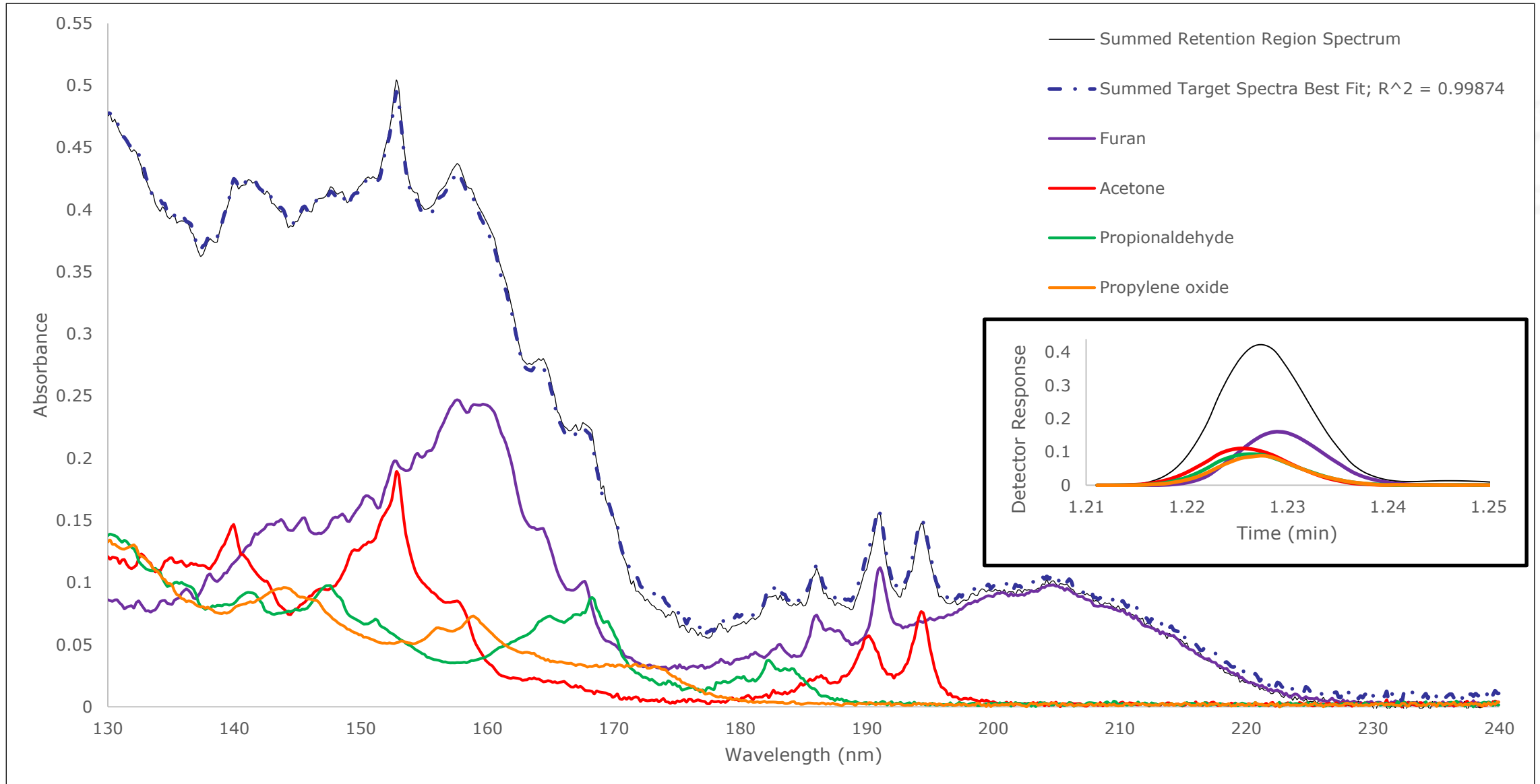
GC Oven Isothermal 100° C



GC Oven Isothermal 200° C



Summed and Individual VUV Absorbance Spectra



Residual Solvents in Pharmaceuticals

Residual Solvents in pharmaceuticals are defined as volatile organic compounds used in or generated from the manufacture of drug substances, pharmaceutical additives, or drug products



Pharmaceutical Regulatory Agencies

- A scientific organization that sets standards for the identity, strength, quality, and purity of medicines, food ingredients, and dietary supplements manufactured and consumed worldwide.
- USP's drug standards are enforceable in the United States by the Food and Drug Administration
- The International Council for Harmonisation (ICH) of Technical Requirements for Pharmaceuticals mission is to achieve greater harmonisation worldwide to ensure that safe, effective, and high quality medicines are developed and registered in the most resource-efficient manner.



Global Expertise
Trusted Standards
Improved Health



ICH

harmonisation for better health



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USP Static Headspace Conditions

Pharmaceutical product for analysis
dissolved in water or another suitable
solvent in a headspace vial.

Headspace Operating Parameter Sets

	1	2	3
Equilibration temperature (°)	80	105	80
Equilibration time (min.)	60	45	45
Transfer-line temperature (°)	85	110	105
Carrier gas: nitrogen or helium at an appropriate pressure			
Pressurization time (s)	30	30	30
Injection volume (mL)	1	1	1

https://www.usp.org/sites/default/files/usp_pdf/EN/USPNF/generalChapter467Current.pdf

USP 467 GC Conditions

- Flame ionization detector (FID) 250°C
 - GC inlet 140°C, split ratio 5
 - 30m x 0.32mm x 1.80µm G43 (624- or 1301-type phase)
 - Or... 30m x 0.53mm x 3.00µm G43
 - Nitrogen or helium 35 cm/sec
 - GC oven: 40°C (20 min), 10°C/min to 240°C (20 min)
-
- Run time: 60 min

https://www.usp.org/sites/default/files/usp_pdf/EN/USPNF/generalChapter467Current.pdf

Can GC—VUV Do Better?

- **Absorbance spectra to identify solvent**
 - Flame ionization detector is non-specific
- **Combine Class 1 and Class 2 solvents in one run**
 - Good detectability and linear range for VUV
- **Faster GC runs for higher sample throughput**
 - Higher GC flow rates which may be unsuitable for other detection platforms
- **Spectral filters provide specificity**
 - 170-240 nm for benzene, for example
- **Spectral deconvolution for coeluting solvents**
 - VUV Vision



GC—VUV Experimental Conditions

VUV Analytics VGA-100

- Transfer line & Flow Cell temperature: 275°C
- Makeup gas pressure: ~0.36 psi
 - GC column flow is 4 mL/min
 - So really just represents flow cell pressure
- Acquisition range: 125 to 240 nm
- Acquisition rate: 4.5 Hz

GC Conditions

- Agilent 6890 GC
- 30 m x 0.25 mm x 1.40 µm Rxi-624Sil MS GC Column
- GC oven: 35°C (1 min), 30°C/min to 245°C
- Run time: 8 min

Gerstel MPS2

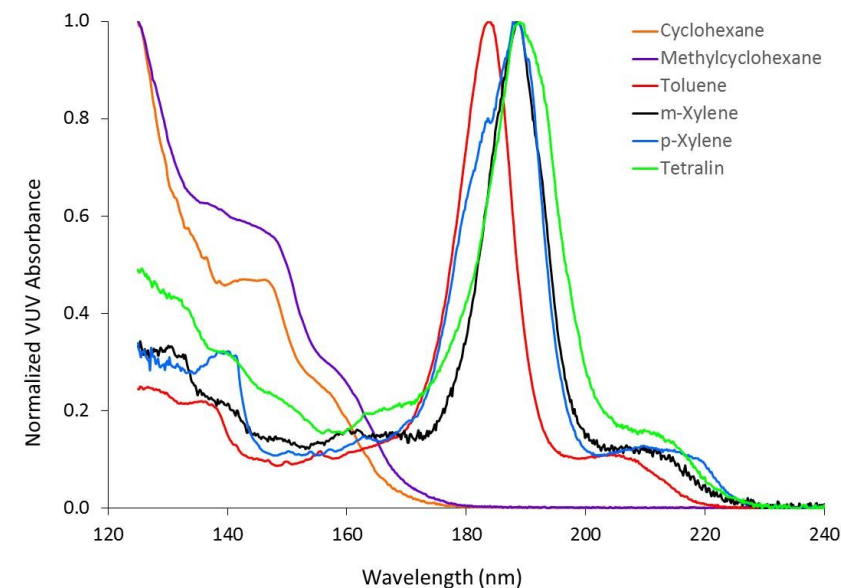
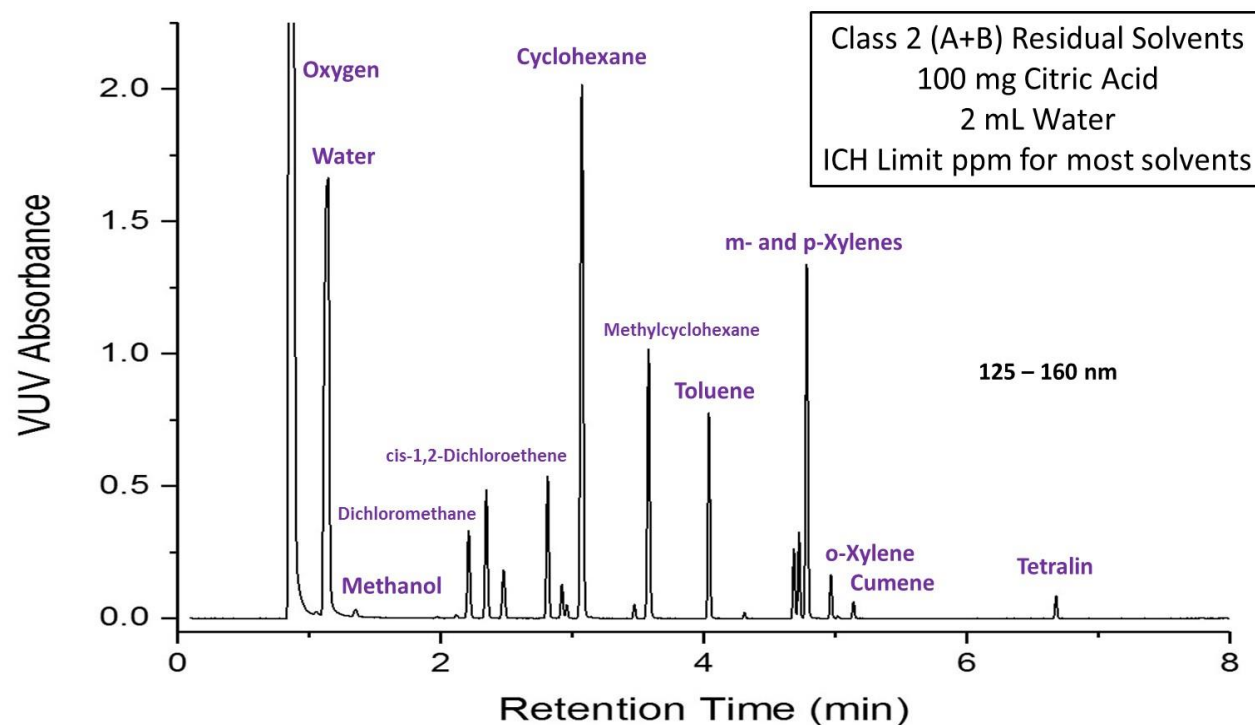
- Syringe temperature: 90°C
- Incubation temperature: 80°C
- Incubation time: 10 min
- Agitator: 250 rpm
 - 10 sec on, 1 sec off
- Injection volume: 250 µL
- Injection speed: 200 µL/se



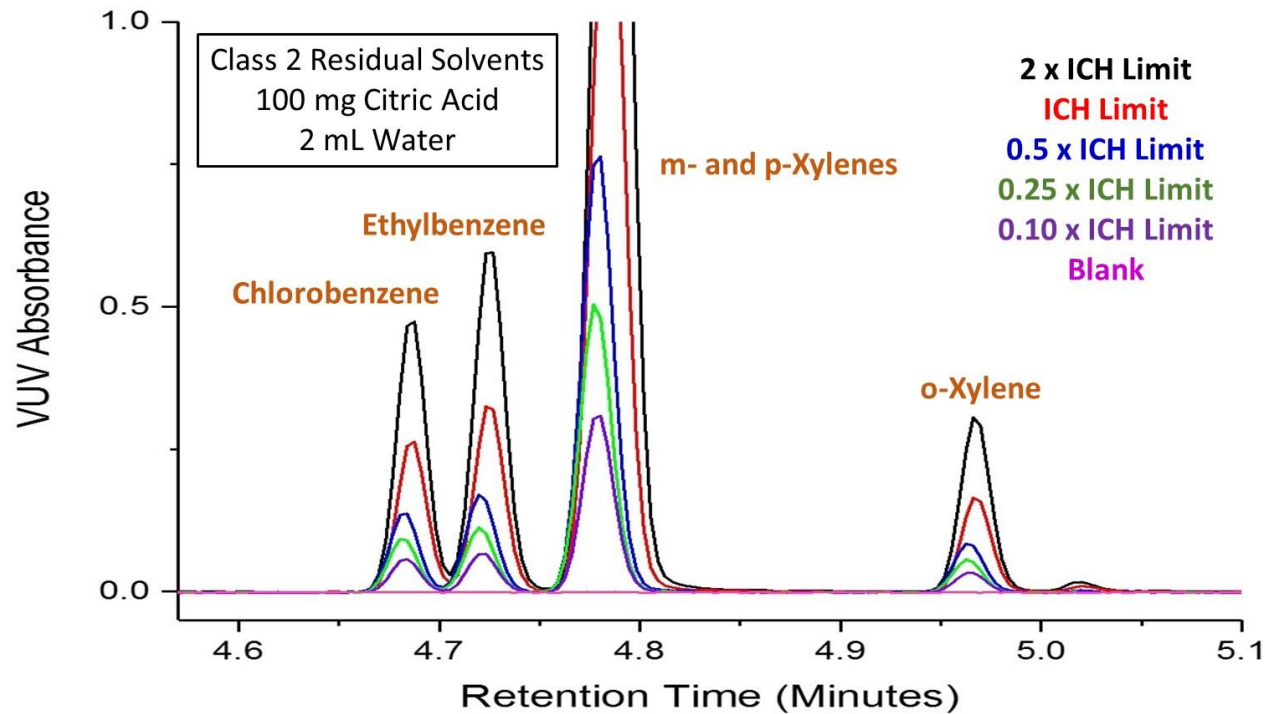
GC conditions

- Agilent 6890 GC
- Topaz 2 mm ID Straight Inlet Liner (Restek)
- 250°C
- Split ratio 2.5
- 30m x 0.25mm x 1.40µm Rxi-624Sil MS GC Column (Restek)
- Helium 4 mL/min (constant flow)
- GC oven: 35°C (1 min), 30°C/min to 245°C
- Run time: 8 min

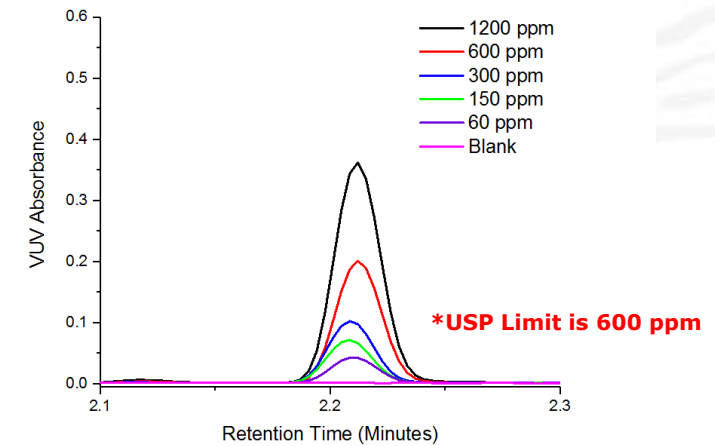
GC-VUV Analysis of Class 2 Residual Solvents



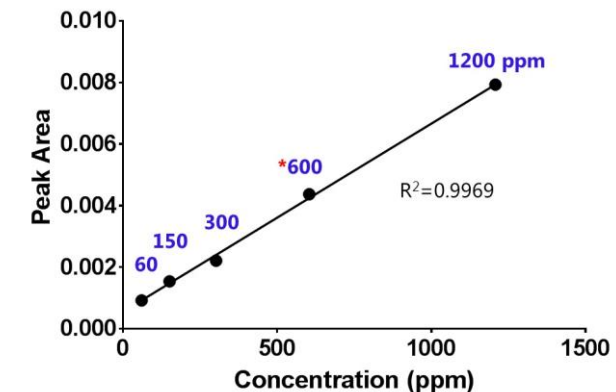
Chromatographic Linearity Above & Below ICH Limits



Dichloromethane

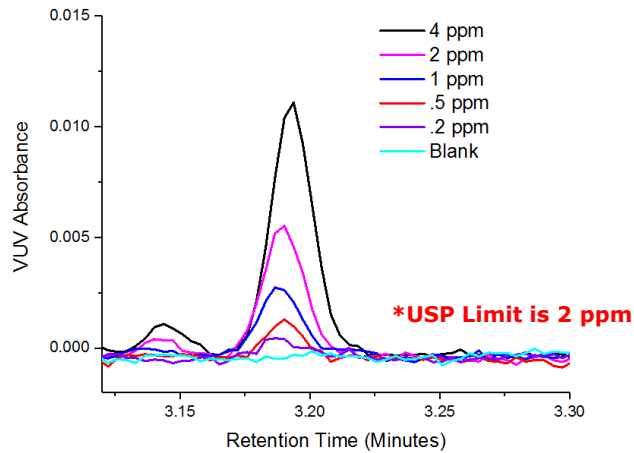


DCM Static Headspace Linearity

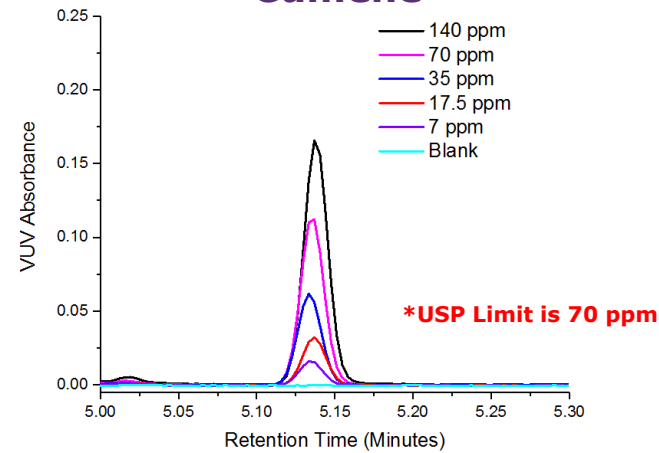


Linearity of Class 1 & 2 Solvent Mix Spiked into Throat Spray

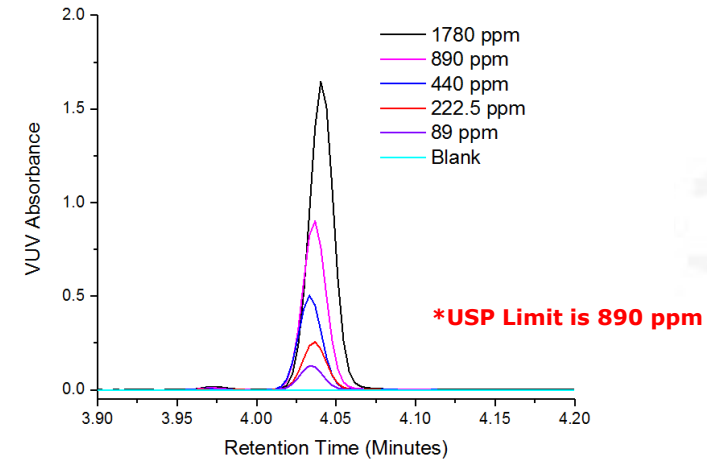
Benzene



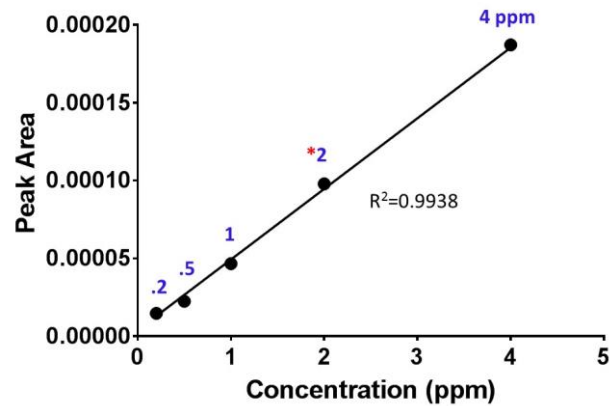
Cumene



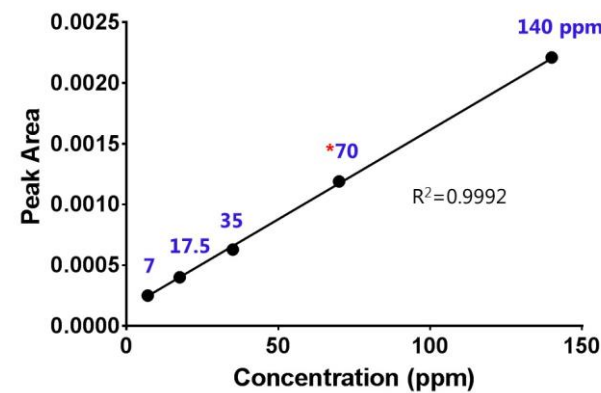
Toluene



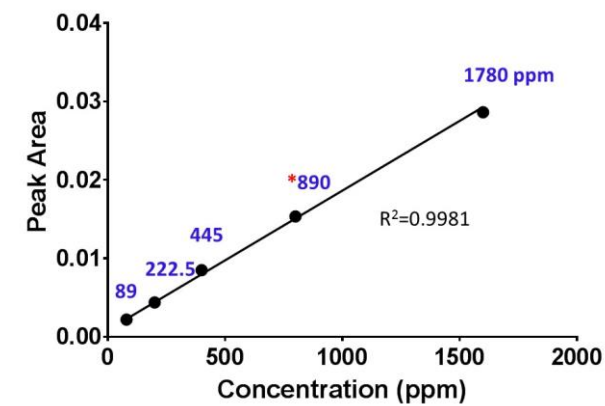
Benzene Static Headspace Linearity



Cumene Static Headspace Linearity

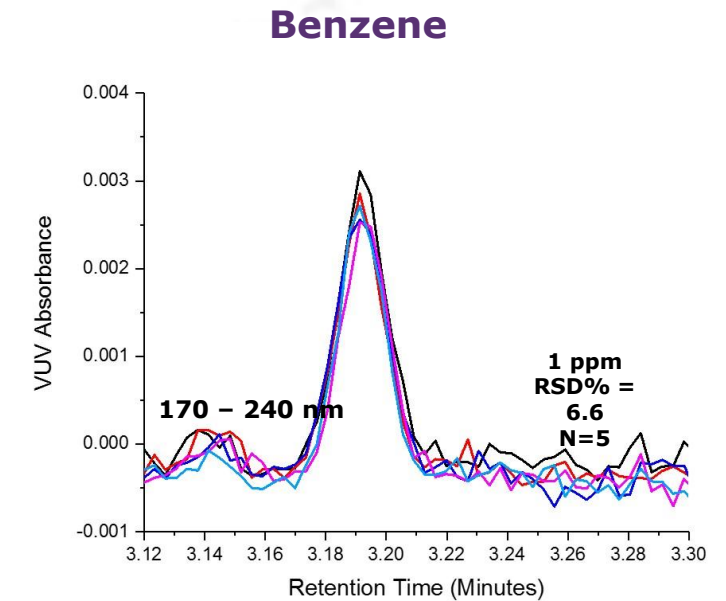
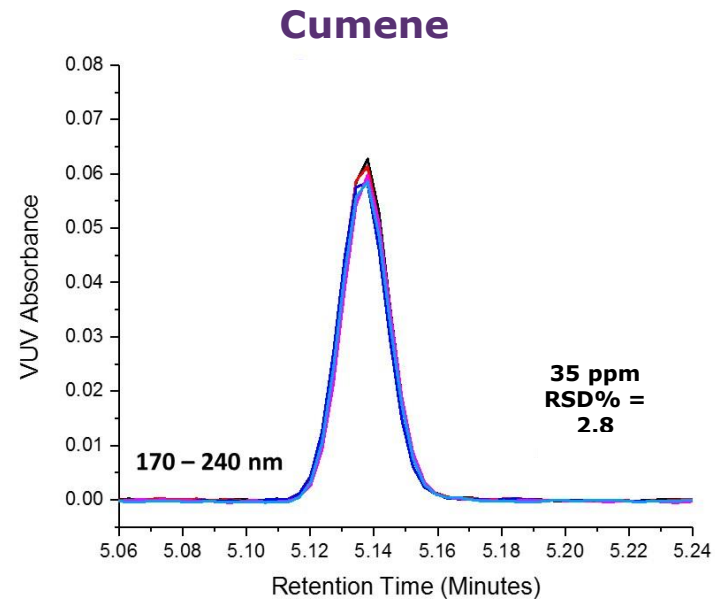
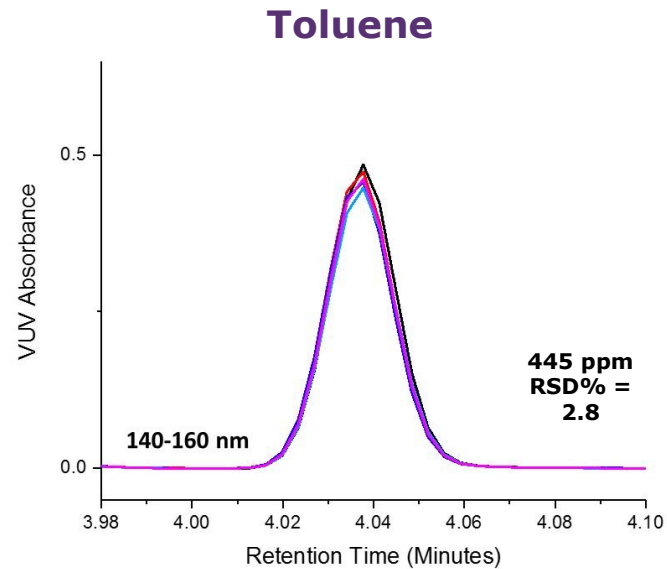


Toluene Static Headspace Linearity



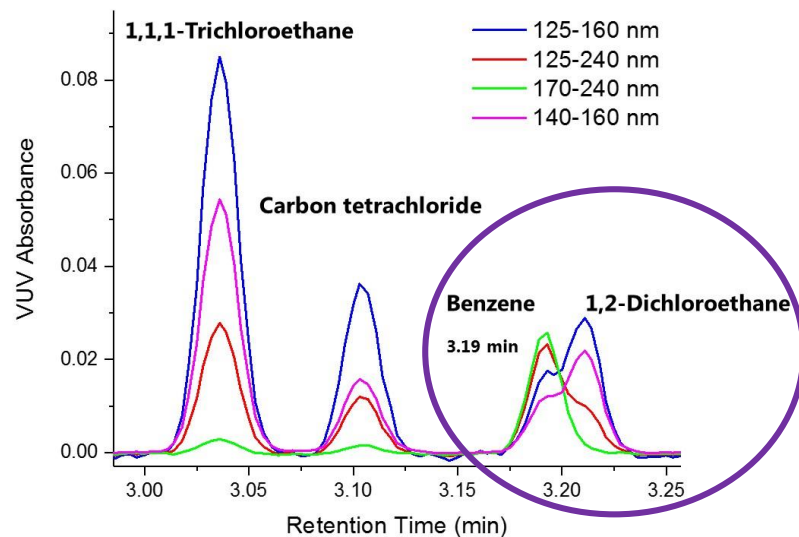
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Reproducibility of Class 1 & 2 Solvent Mix Spiked to Throat Spray

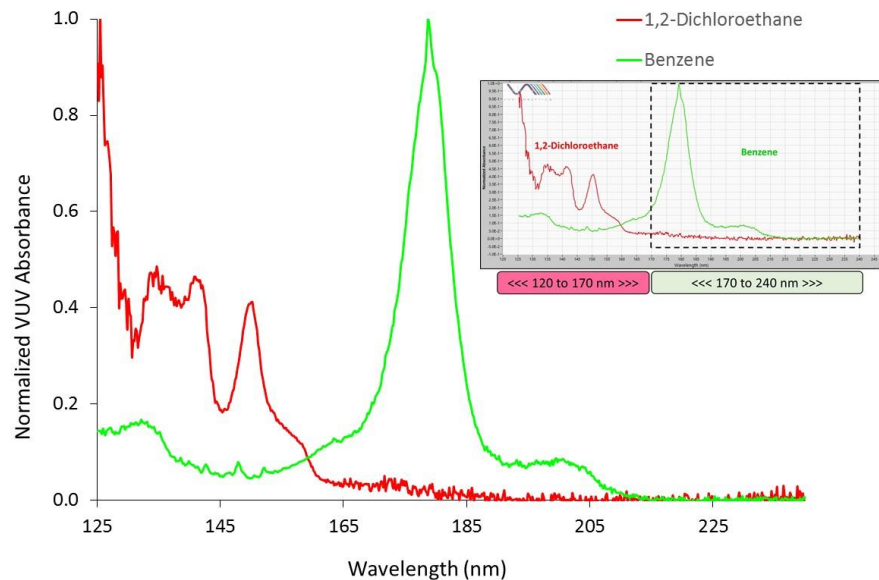


Co-elution & Deconvolution of Class 1 Residual Solvents

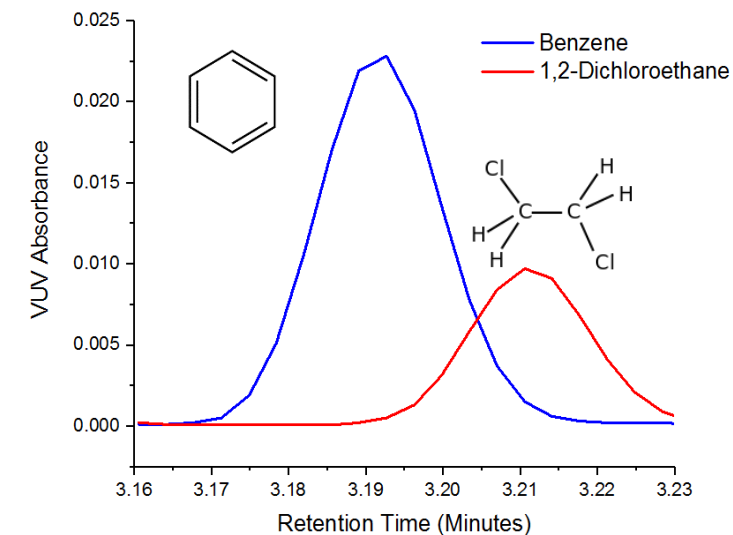
Chromatogram Zoomed-in on Class 1 Co-Elution



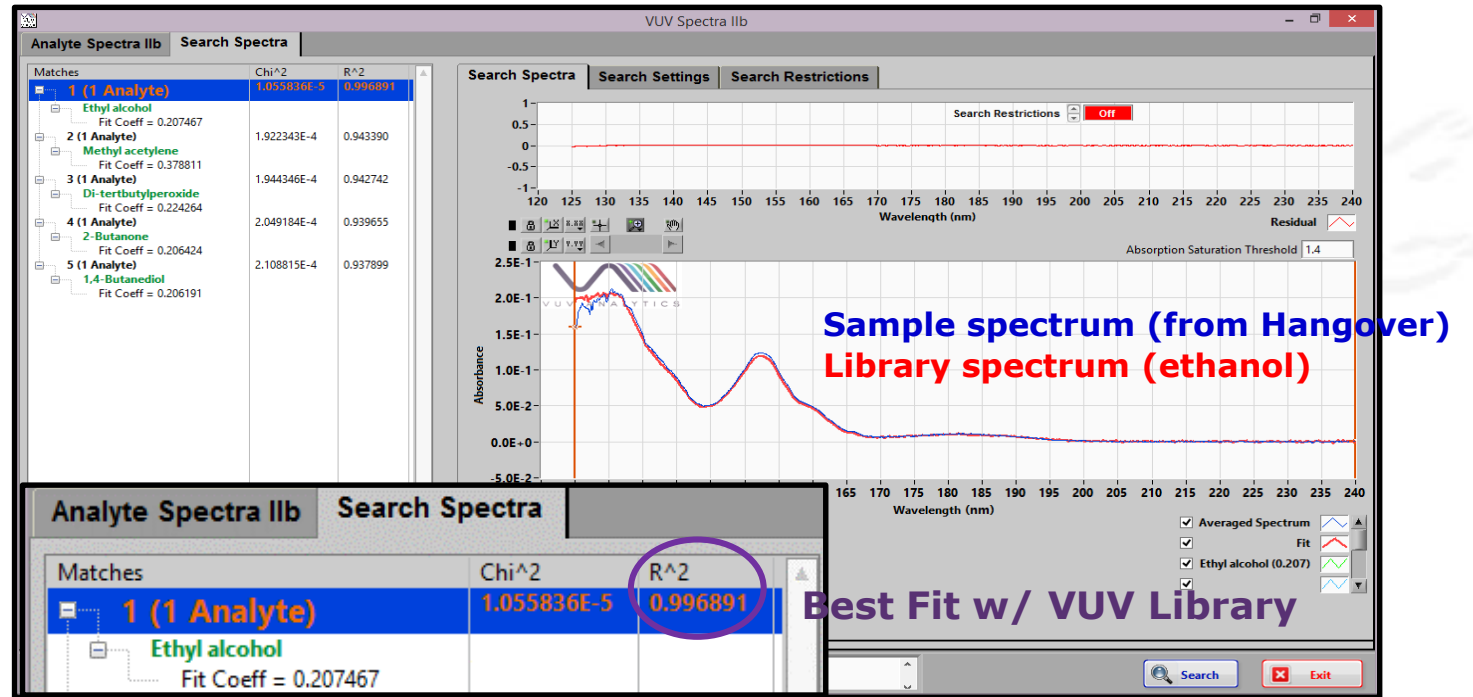
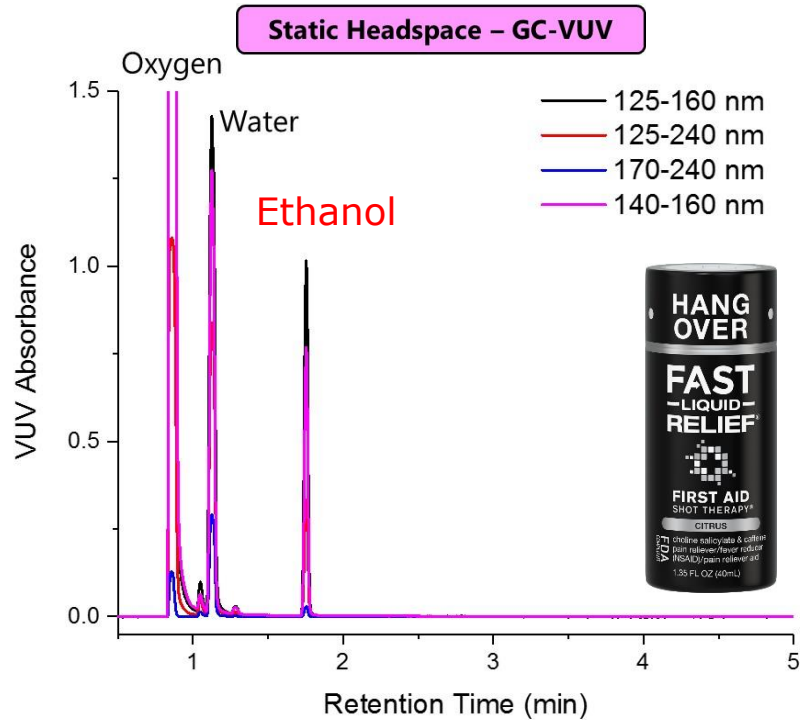
Spectral Comparison of Co-Eluting Analytes



Deconvolution of Co-Eluting Analytes



Analyzing Untargeted Unknowns by GC-VUV: Hangover Pain Relief



GC—VUV Key-features

- Universal and selective detection
- Unique, class similar spectra
- Complementarity to MS
- Deconvolution of coeluting analytes
- Automated classification and speciation of mixtures
- Good quantitative performance
- Theoretical computations



Acknowledgements

- Jack Cochran – VUV Analytics
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- James Diekman – VUV Analytics
- Lindsey Shear – VUV Analytics
- Kevin Schug – University of Texas at Arlington



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