

Reduction of analysis times in HPLC at elevated column temperatures

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Contents of this lecture

- Four major practical questions
- Sample throughput, Analysis time reduction
- Conclusions / remarks

Four major practical questions

- How to select the optimal column for a specific separation from the several hundreds available, and in many cases, nominally identical, RP-columns?
- How to perform an objective and fast method development procedure resulting in an optimal and rugged analysis protocol, Sample throughput?
- Upon validation of an RPLC analysis protocol, what will be the repeatability and reproducibility of columns and batches of a specific stationary phase guaranteeing an undisturbed continuous analysis process.
- Once column and separation method have been worked out, what will be the longevity of that column under the applied conditions; In other words, how many sample injections can be performed in a column life cycle time.

Analysis time

Why faster analysis ?

- Faster QC, process control and other analysis
- Handle more complex samples without increasing analysis time
- Rapid method development
- Reduce solvent use/disposal
- Reduce costs
- Improves total productivity

Analysis time

Analysis speed can be substantially improved

- Ultra high pressure liquid chromatography (UPLC)
- Capillary electro chromatography (CEC)
- Monolithic columns
- Increasing column temperature

Analysis time

Requirements:

- Columns of high permeability
- Short packed columns: 50 mm or shorter; Small particle sizes; 3.5 μm or smaller: 1.8 μm
- Monolithic columns
- Optimized HPLC equipment; eliminating extra-column band broadening effects; tubing, detector cell, pumps delivering increased pressure and/or eluent flows, well designed thermostating systems
- Electronics; detector response time – data sampling rate

Analysis time

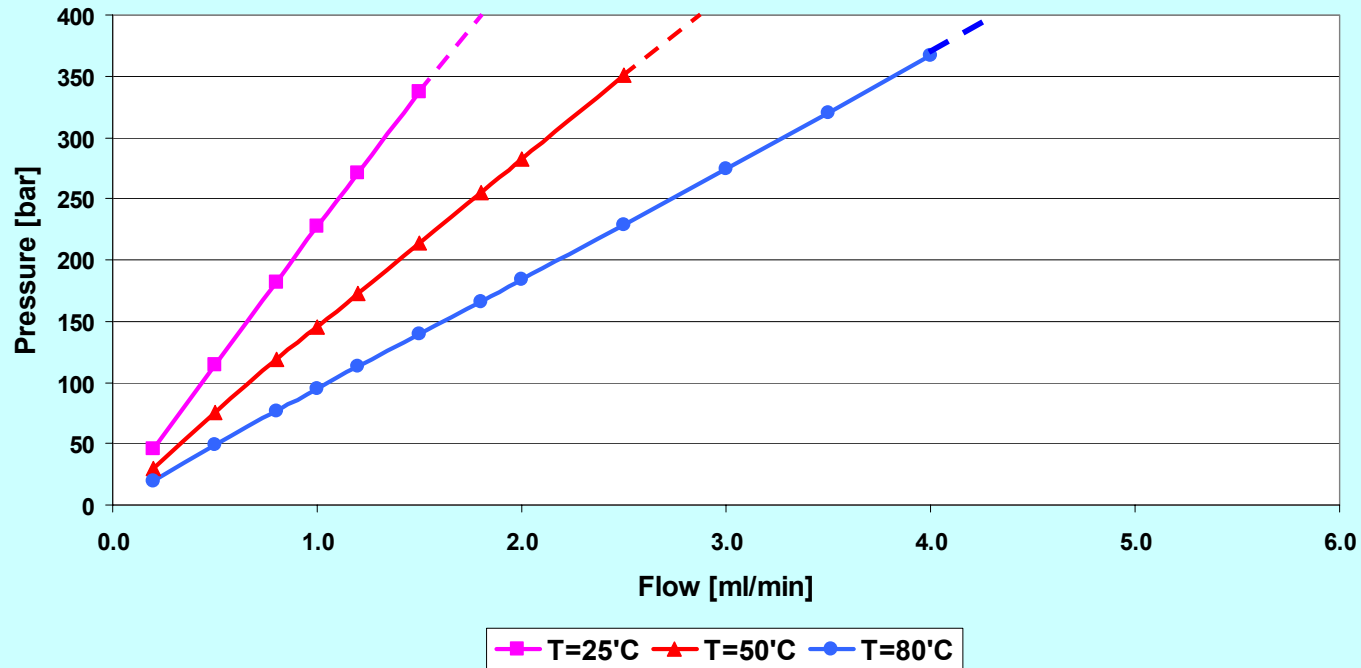
Analysis at increased temperature require:

- thermally stable stationary phases
- thermally stable analytes
- optimized HPLC equipment
- knowledge of influence T on retention and selectivity

Analysis time

dp = 1.8 μ m

Eluent: Methanol/Water = 60/40 (v/v)



Analysis time

Column: SB-C18, 4.6 x 50 mm; 1.8 μm

$$\frac{dF}{dP} = (\text{ml} / 100\text{bar} \cdot \text{min})$$

Eluent: Methanol / Water = 60/40 (v/v)

25°C	50°C	80°C
0.45	0.72 (1.6)	1.10 (2.4)

Eluent: Acetonitrile / Water = 50/50 (v/v)

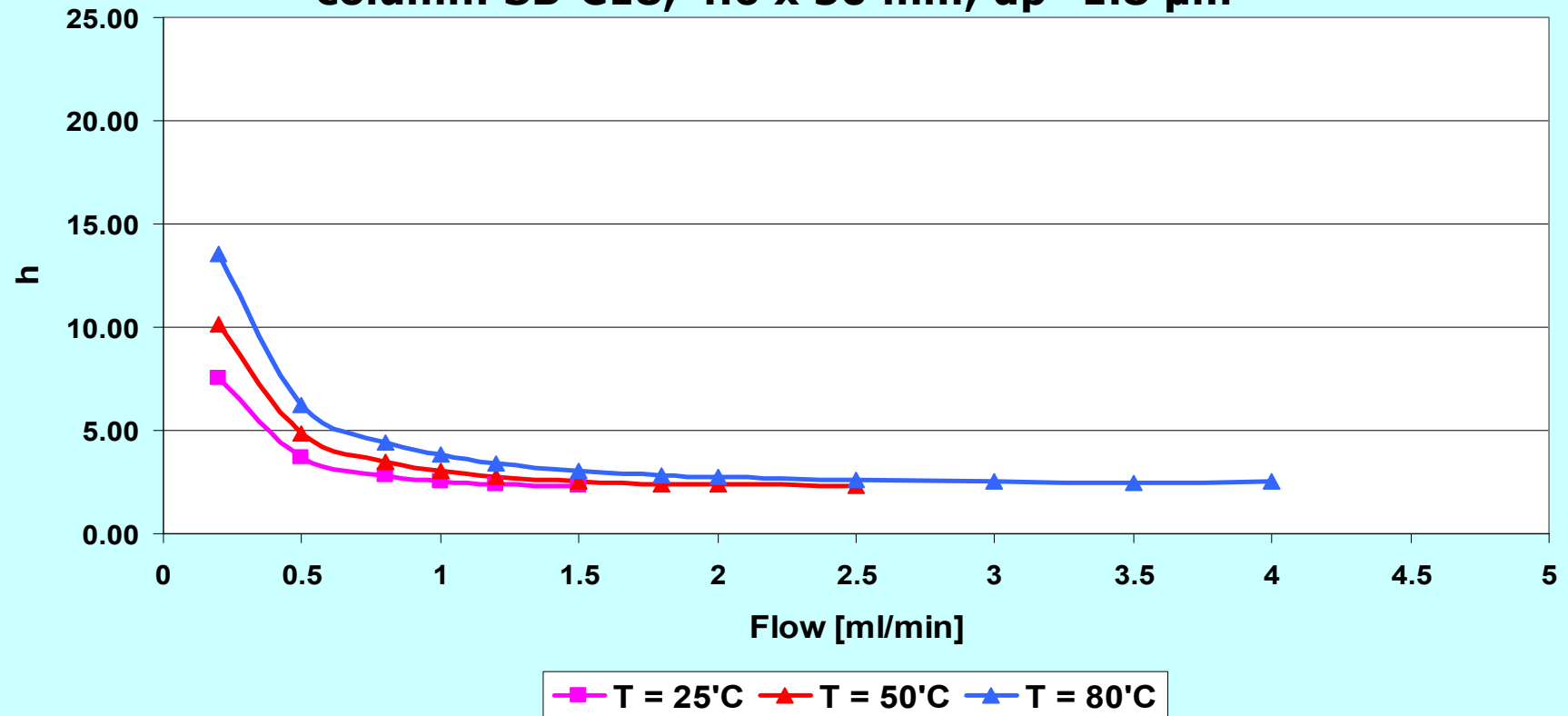
25°C	50°C	80°C
0.77	1.11 (1.4)	1.52 (1.9)

Analysis time

h-flow curves [calculated from N Gauss]

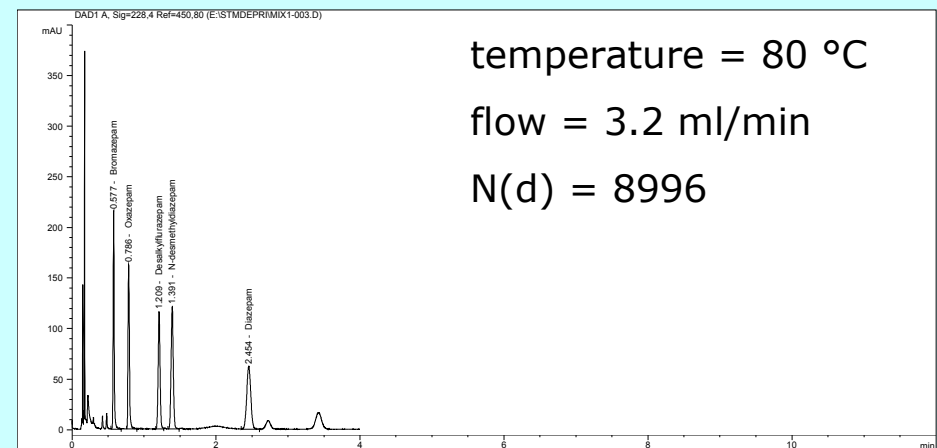
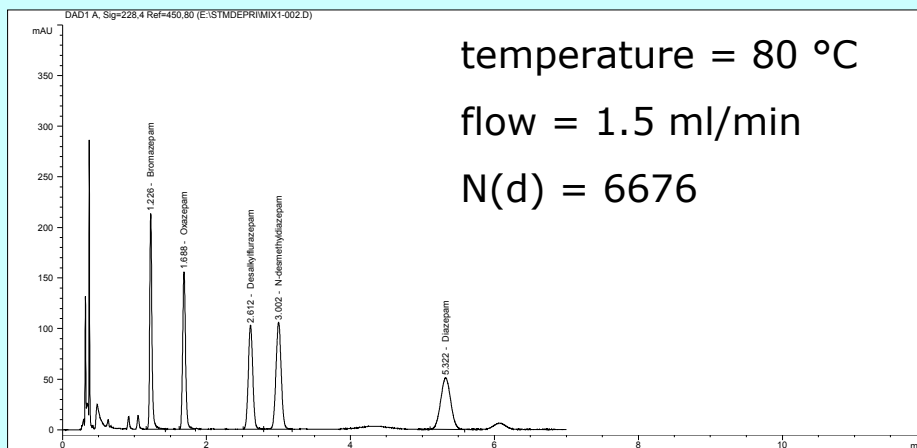
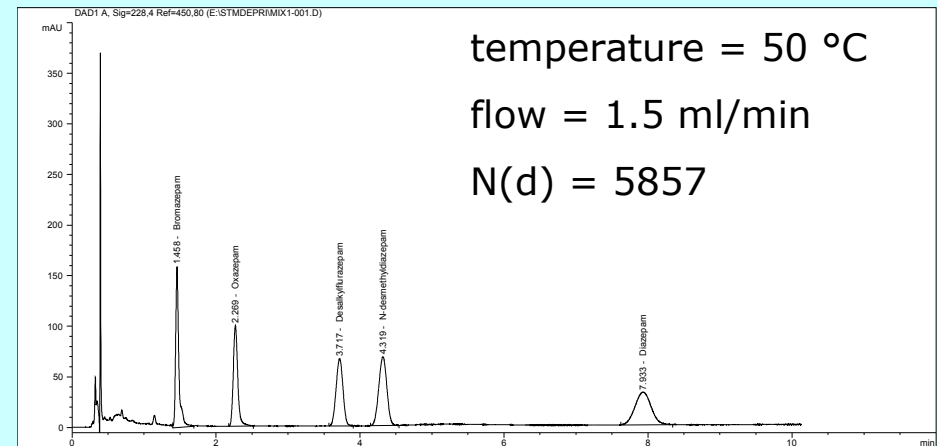
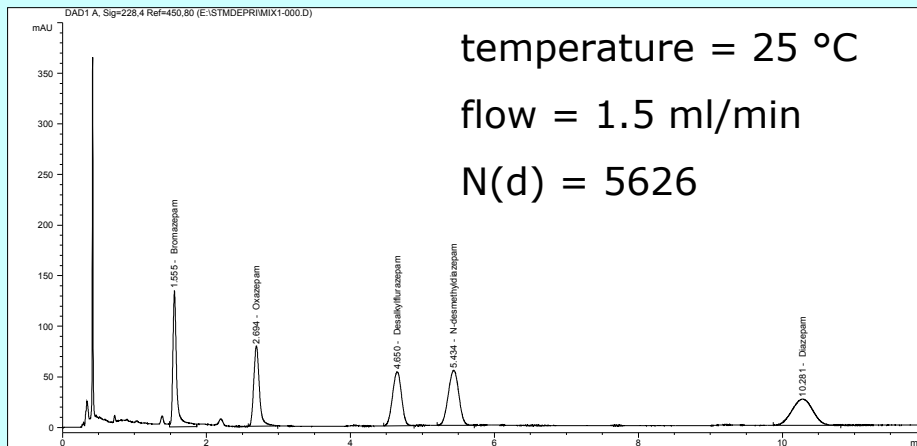
Methanol / Water = 60/40 (v/v)

column: SB-C18; 4.6 x 50 mm; dp=1.8 μ m



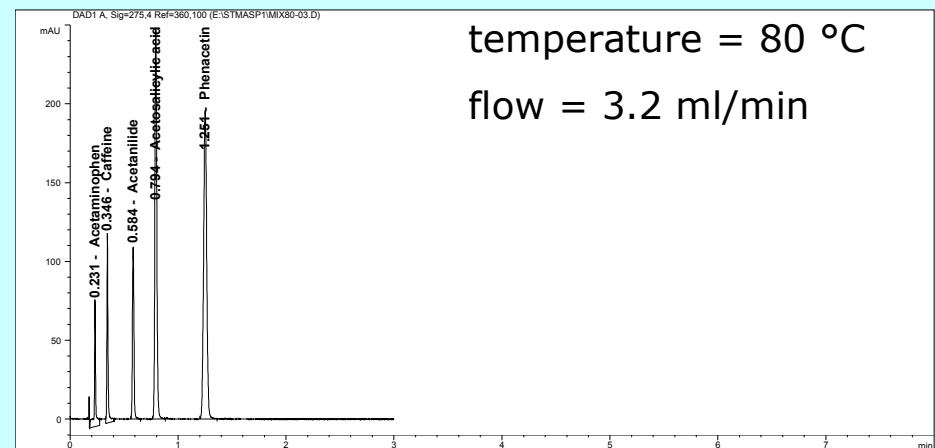
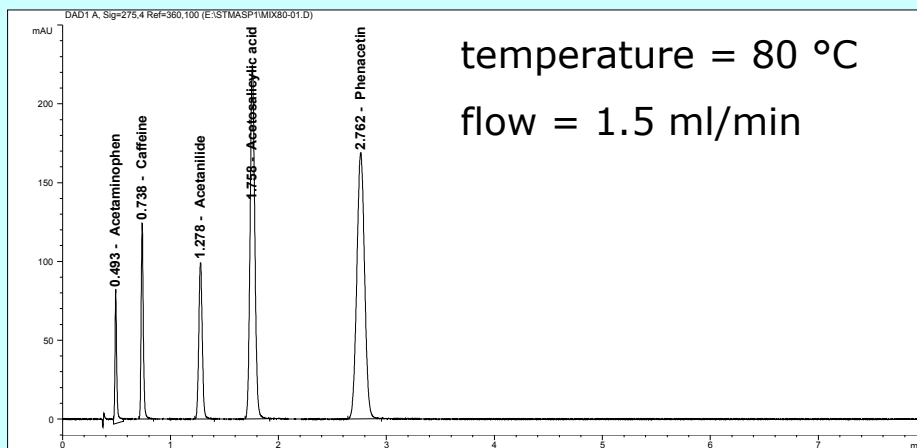
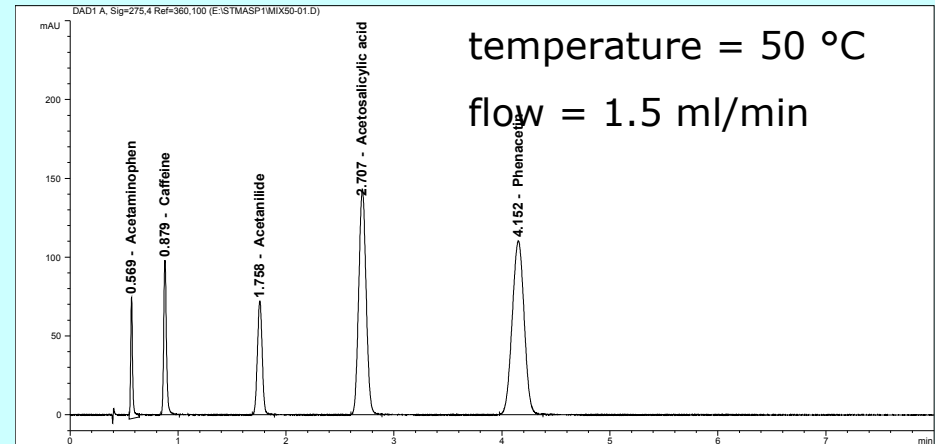
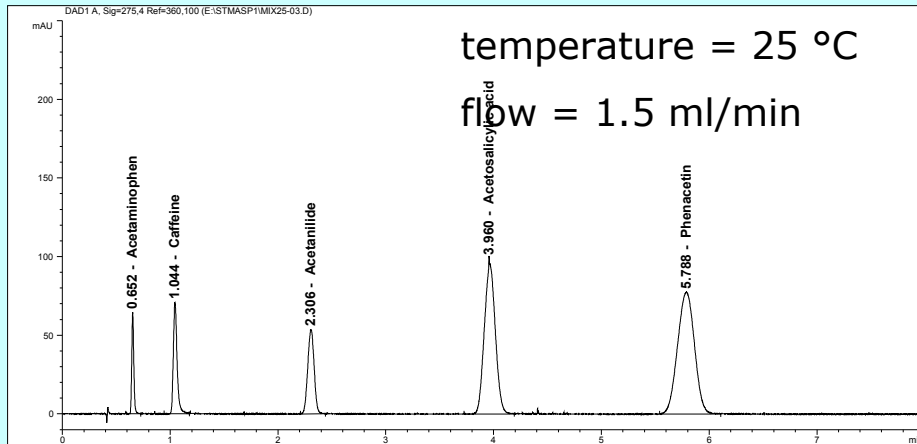
Application examples

Separation of Sedativa



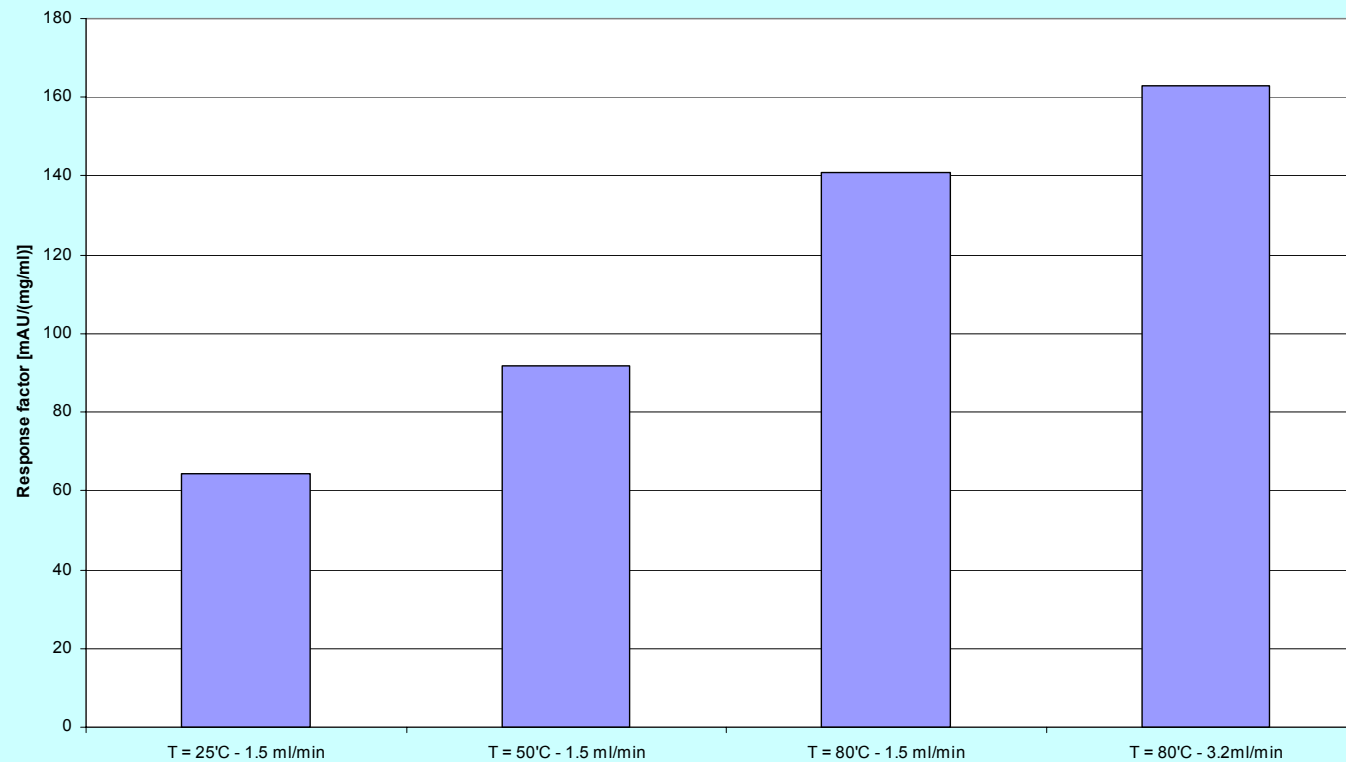
Application examples

Separation of Analgetics



Application examples

Response factor for Phenacetin



Conclusions / remarks

- Upon temperature increase combined effects of cubical liquid expansion, decreasing eluent viscosity and increased analytes diffusion are involved
- Up to 80 °C analysis time can be reduced by a factor of about 4; conventional equipment
- Requirements: columns and analytes (dynamic) must be thermally stable

Conclusions / remarks

- Chemically and thermally stable RPLC phases based on silica (90 °C); other inorganic oxides (190 °C) and polymers (150 °C) are available
- Apart from UPLC, CEC and monoliths also temperature is a strong tool to reduce analysis times

Acknowledgement

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