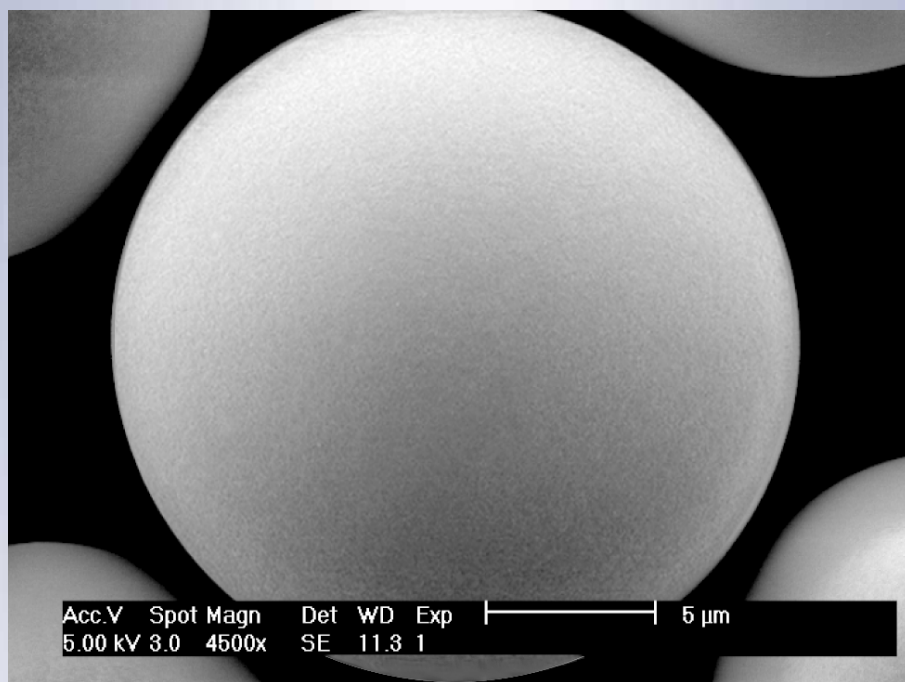

**Criteria for the selection of a
suitable RP phase in HPLC,
showed with NUCLEODUR[®] examples**

Stationary phase selection

- Silica support
 - ✓ **Mechanical properties/metal contaminants**
- Surface chemistry
 - Base deactivation
 - Retention and selectivity
 - pH-stability and selectivity
 - Stability in 100% aqueous eluents
 - Special selectivity features

Demands on high performance RP silicas



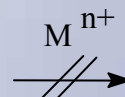
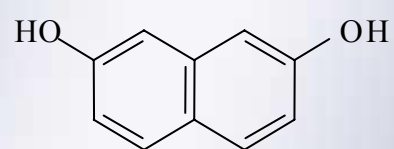
NUCLEODUR 100 Å

- **high level of purity**

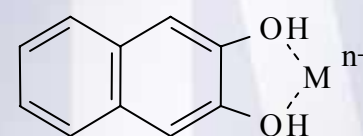
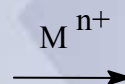
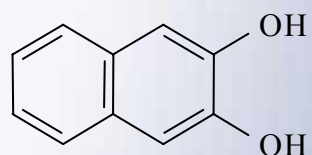
Al	< 5	ppm
Fe	< 5	ppm
Na	< 5	ppm
Ca	< 5	ppm
Ti	< 1	ppm
Zr	< 1	ppm
As	< 0.5	ppm
Hg	< 0.05	ppm

- **homogeneous surface for optimal coverage**

Determination of metal contaminants on the silica surface



2,7-Dihydroxynaphthalene (2,7-Dert)

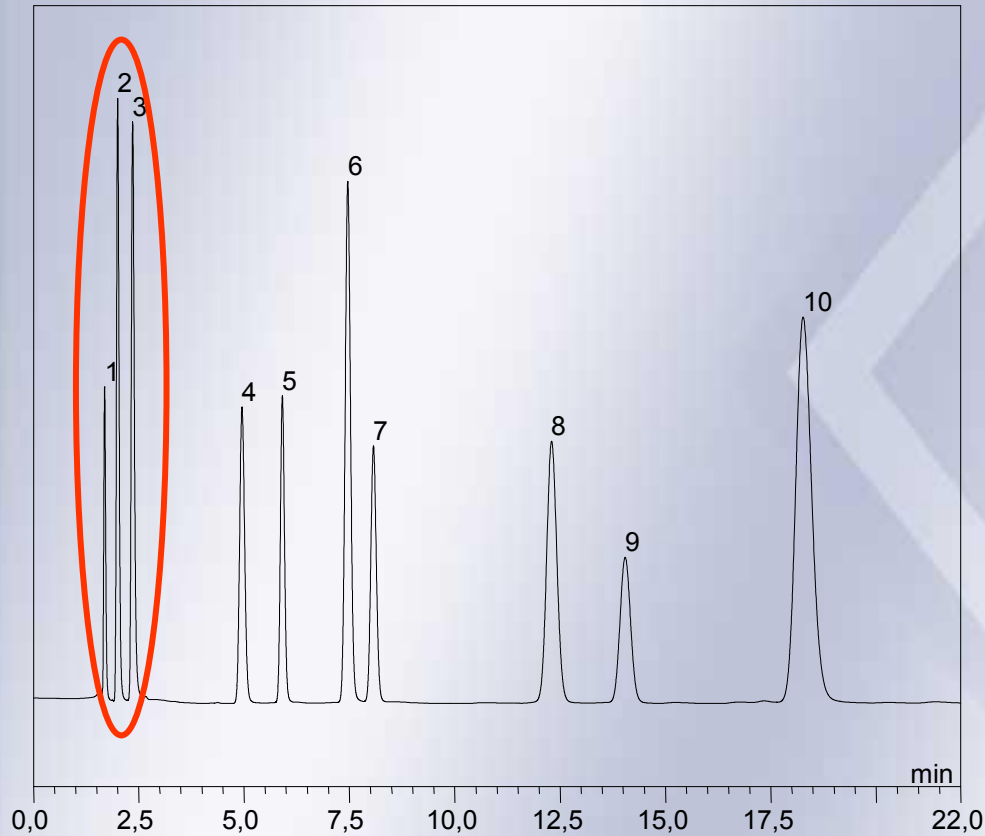


2,3-Dihydroxynaphthalene (2,3-Dert)

broad and
tailing peak

**An indication of metal ions incorporated
on the silica surface is the quotient
 $As(2,3-Dert)/As(2,7-Dert)$**

Determination of metal contaminants on the silica surface



Column: EC 250/4 **Nucleodur** C18 Gravity, 5 μm

Eluent: MeOH/ 20 mM KH_2PO_4 , pH 7,
75:25 (v/v)

Temp.: 30°C

Flow: 1 ml/min

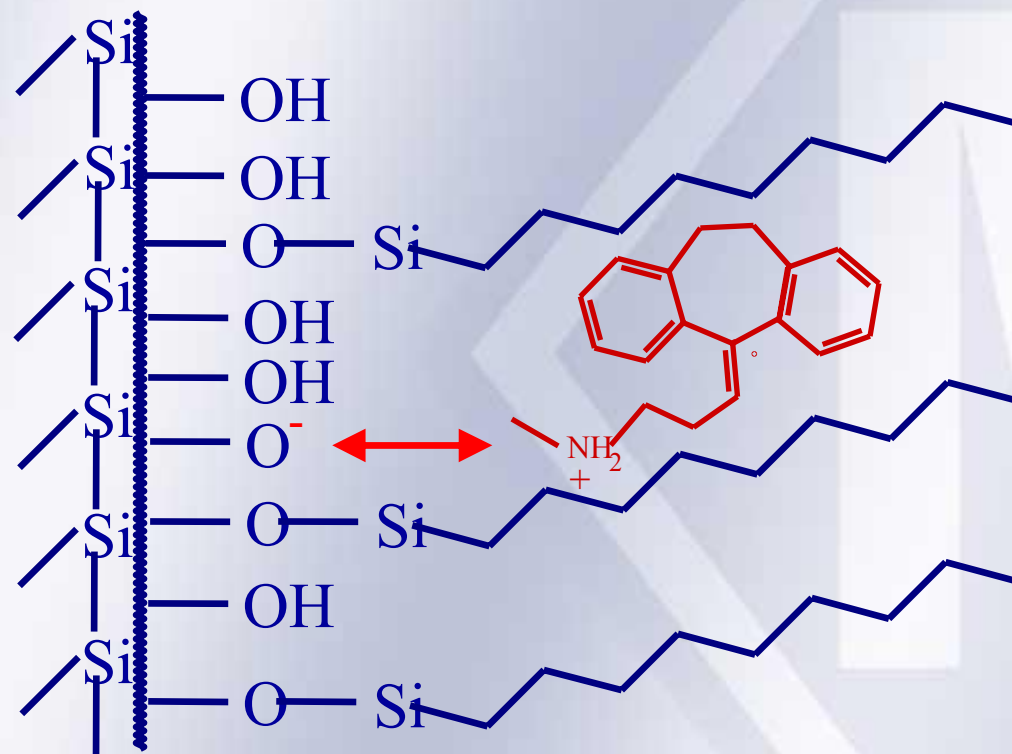
Detection: UV, 254 nm

1. Uracil
2. 2,7-Dihydroxynaphthalene
3. 2,3-Dihydroxynaphthalene
4. Lidocaine
5. Toluene
6. Naphthalene
7. Ethylbenzene
8. Dibutylphthalate
9. Acenaphthene
10. Amitriptyline

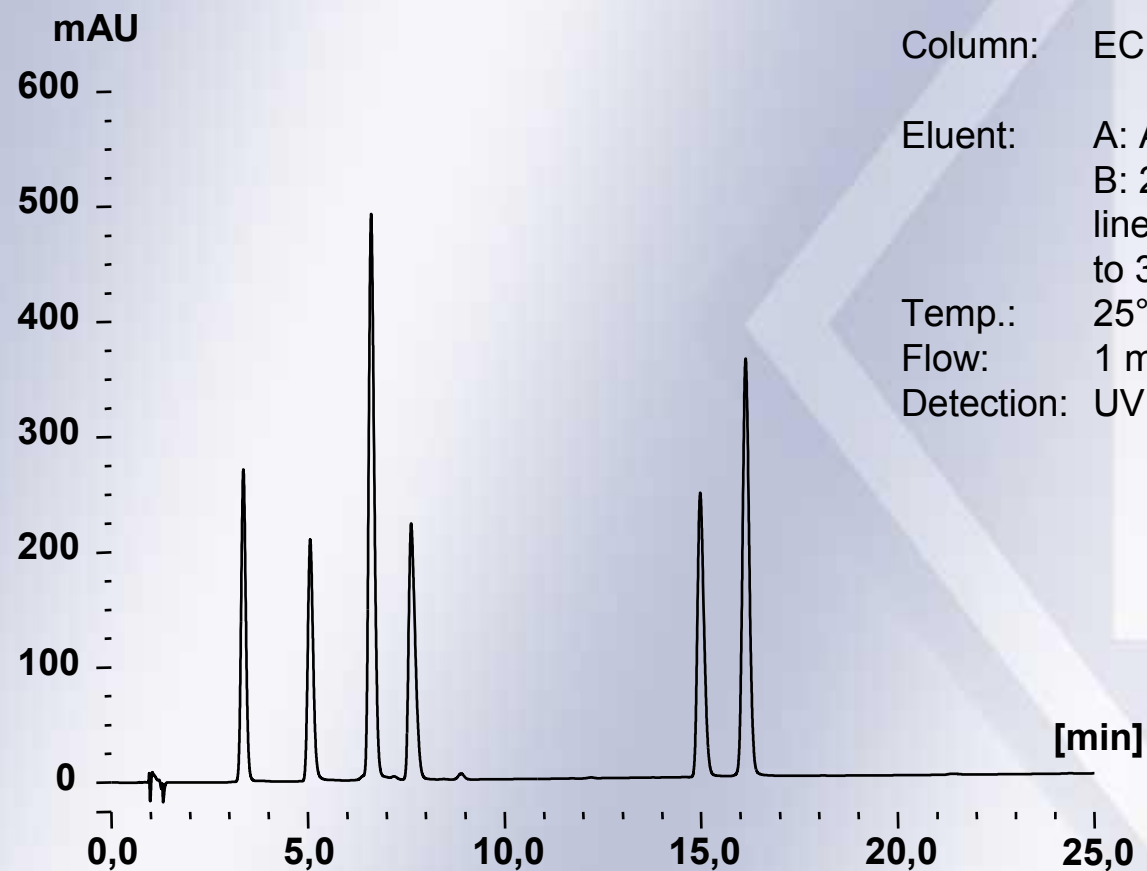
Stationary phase selection

- Silica support
 - ✓ Mechanical properties/metal contaminants
- Surface chemistry
 - ✓ **Base deactivation**
 - Retention and selectivity
 - pH-stability and selectivity
 - Stability in 100% aqueous eluents
 - Special selectivity features

Retention of basic compounds



Alkaloids



Column: EC 125/4 Nucleodur **C18 Gravity**, 5 μm

Eluent: A: ACN
B: 20 mM KH_2PO_4 , pH 2.5,
linear gradient from 10% A in 25 min
to 30% A

Temp.: 25°C

Flow: 1 ml/min

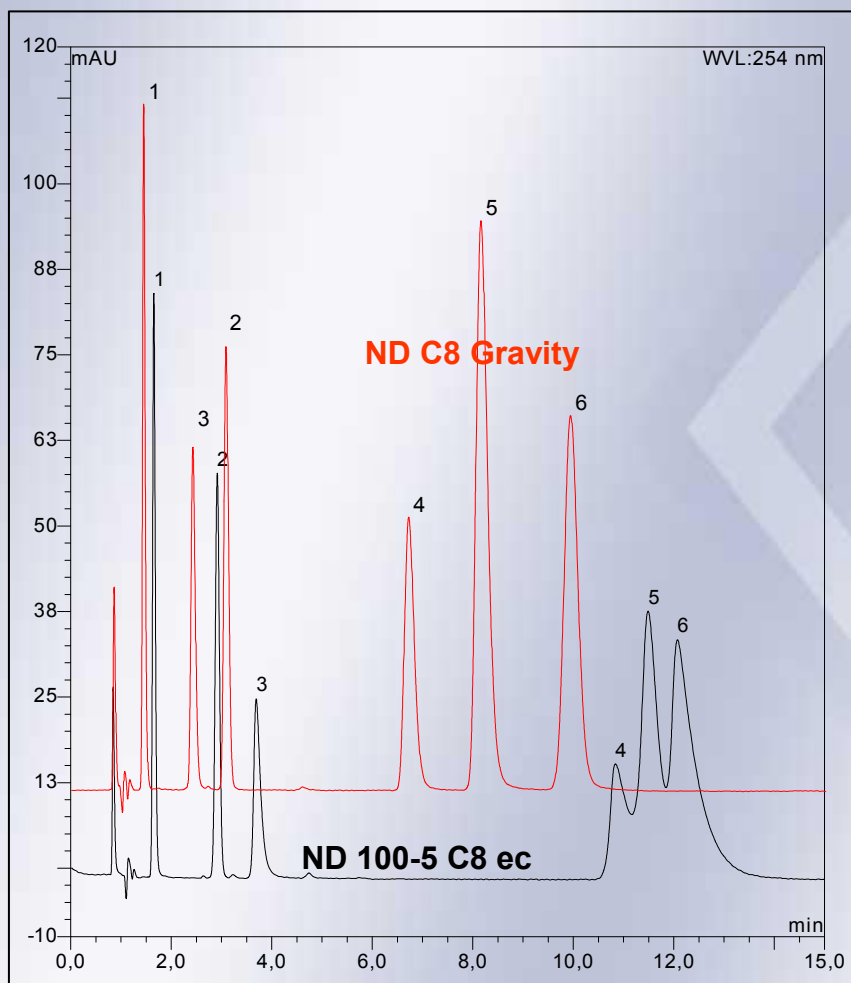
Detection: UV, 210 nm

1. Codeine
2. Quinine
3. Strychnine
4. Atropine
5. Papaverine
6. Noscapine

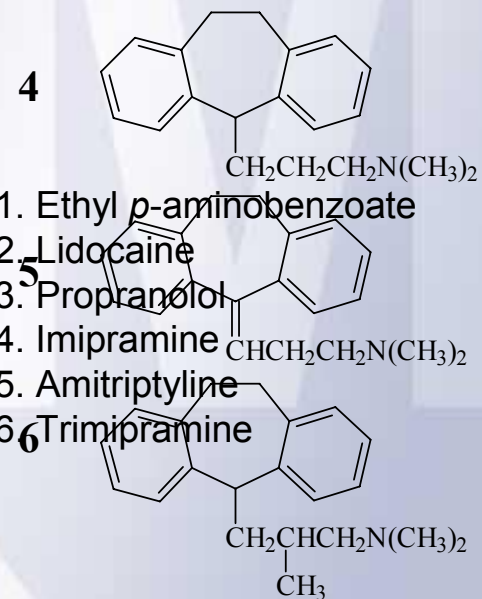
Stationary phase selection

- Silica support
 - ✓ Mechanical properties/metal contaminants
- Surface chemistry
 - ✓ Base deactivation
 - ✓ **Retention and selectivity**
 - pH-stability and selectivity
 - Stability in 100% aqueous eluents
 - Special selectivity features

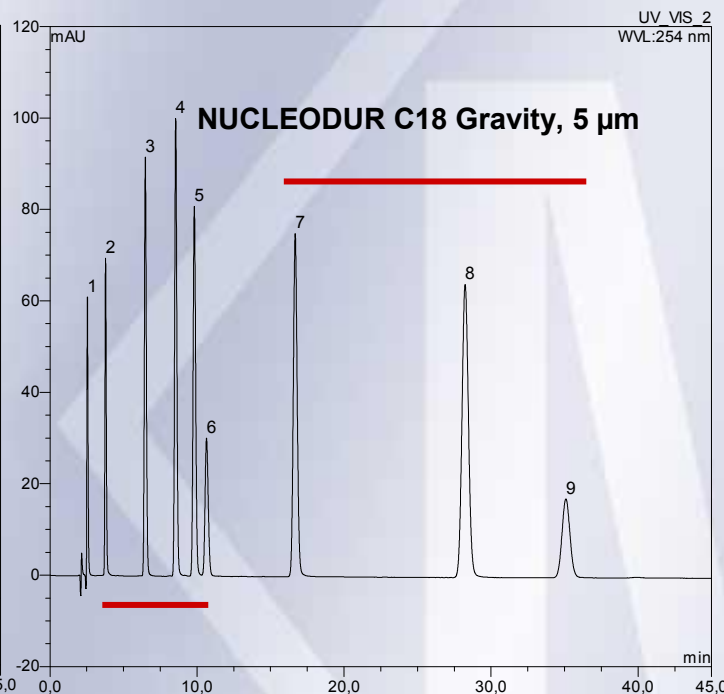
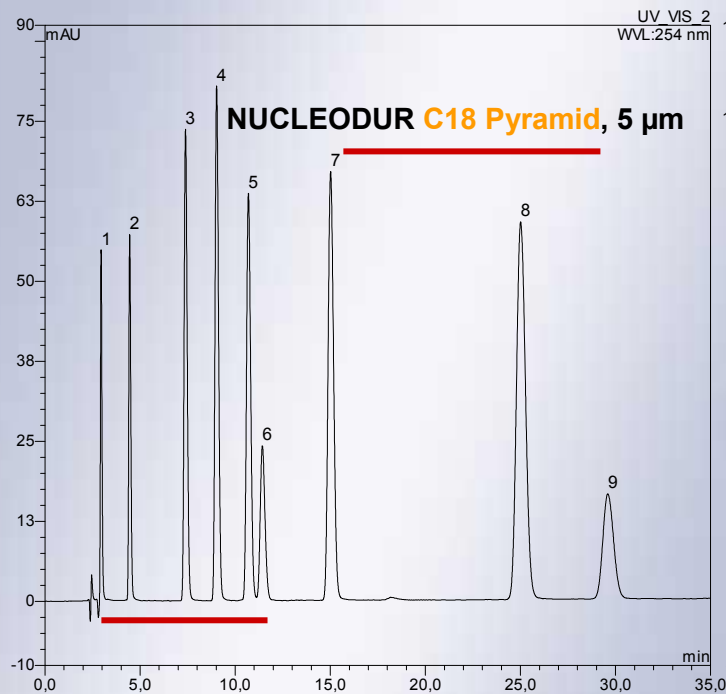
Selectivity of C8 phases with different silylation reagents



Column: **EC 125/4 Nucleodur C8 Gravity, 5 μ m**
 EC 125/4 Nucleodur 100-5 C8 ec
 Eluent: MeOH/ 20 mM KH_2PO_4 , pH 7,
 70:30 (v/v)
 Temp.: 30°C
 Flow: 1 ml/min
 Detection: UV, 254 nm



Polar and non-polar selectivity of C18 phases with different kind of endcapping



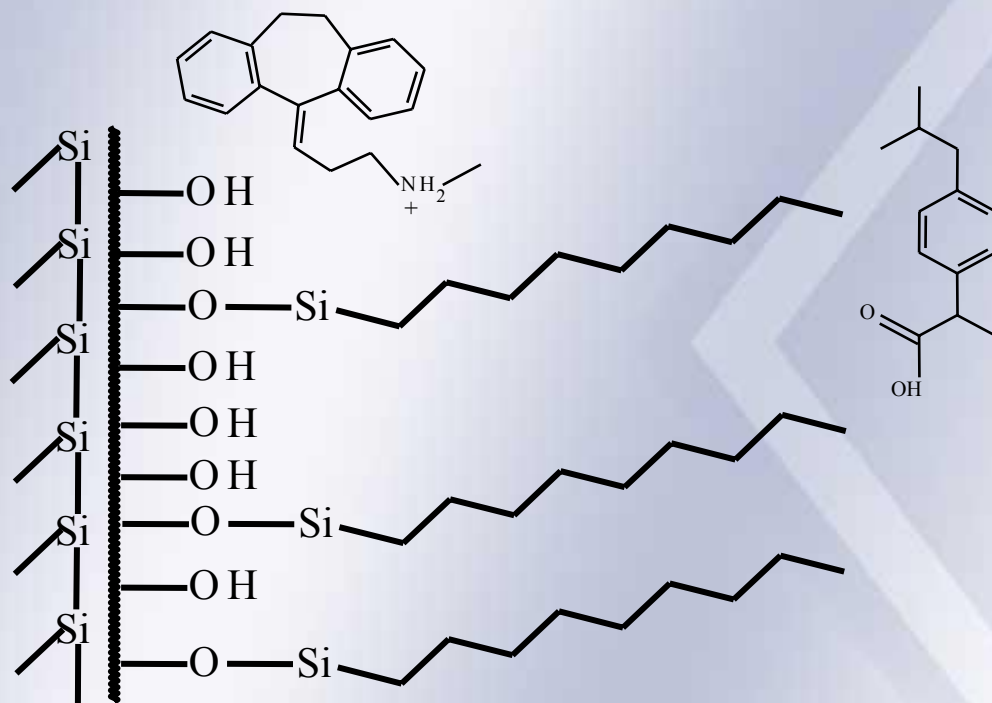
1. Chlorphenyramine
2. Dimethylphthalate
3. Benzamide
4. Ethylbenzoate
5. Benzophenone
6. Lidocaine
7. Naphthalene
8. Biphenyl
9. Acenaphthene

Column: 250 x4 mm
Eluent: MeOH/25 mM $\text{NH}_4\text{H}_2\text{PO}_4$; pH 7
65:35 (v/v)
Flow rate: 0.8 ml/min
Temp.: 40 °C
Detection: UV, 254 nm
Inj. Vol.: 5 µl

Stationary phase selection

- Silica support
 - ✓ Mechanical properties/metal contaminants
- Surface chemistry
 - ✓ Base deactivation
 - ✓ Retention and selectivity
 - ✓ **pH-stability and selectivity**
 - Stability in 100% aqueous eluents
 - Special selectivity features

Retention of ionizable analytes in correlation with pH

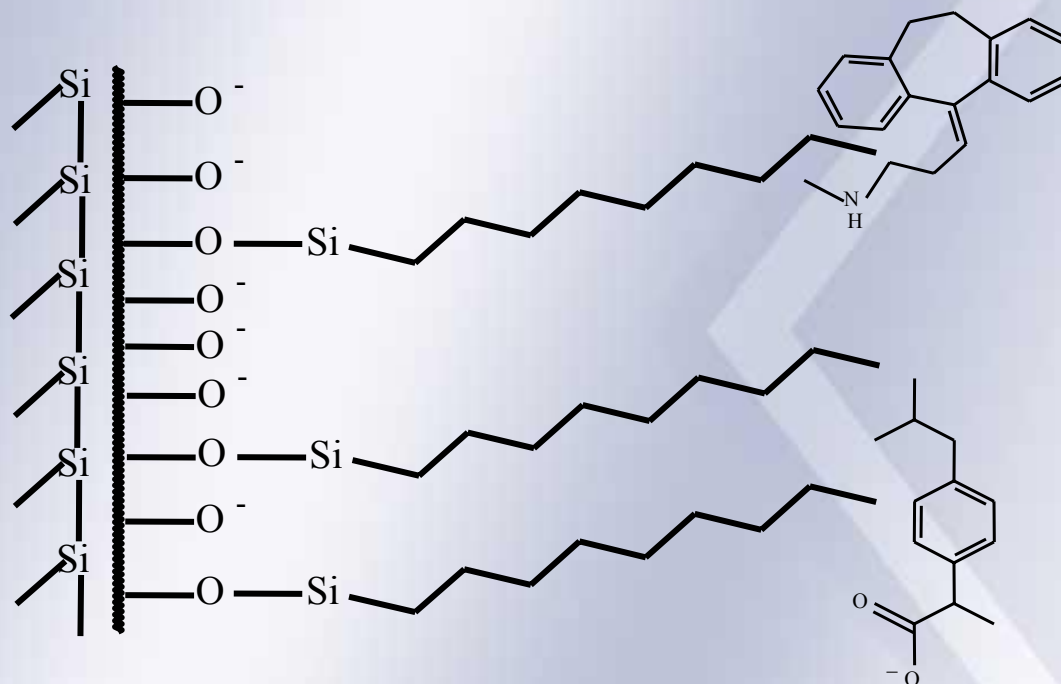


pH \ll 7

- surface silanols protonated
- basic analytes ionized
- poor retention for amino compounds
- acidic analytes neutrally charged
- good retention for acid compounds

-> Cleavage of the siloxane bond by hydrolysis!

Retention of ionizable analytes in correlation with pH

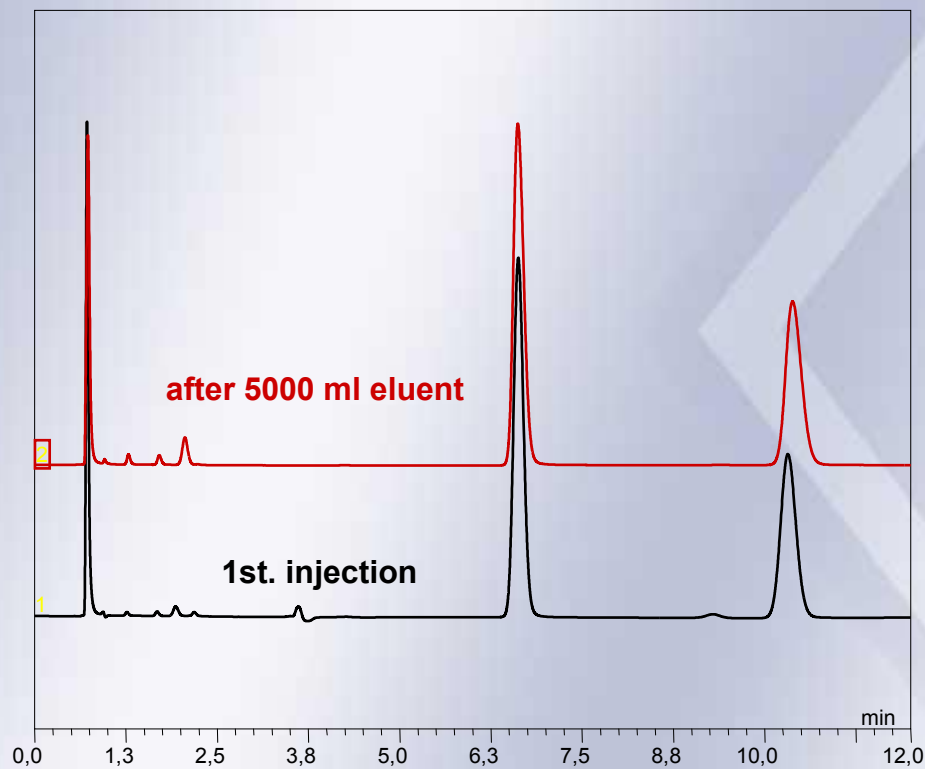


pH >> 7

- deprotonation of surface silanols
- basic compounds neutrally charged
- sufficient retention of amines
- acidic compounds ionized
- poor retention of acids

-> Attack on the silica support and gradual dissolution of the silica!

Stability at acidic pH



Column: EC 125/4 Nucleodur **C18 Gravity**,
5 μm

Eluent: Acetonitrile
1% TFA in water
50:50 (v/v), **pH 1.5**

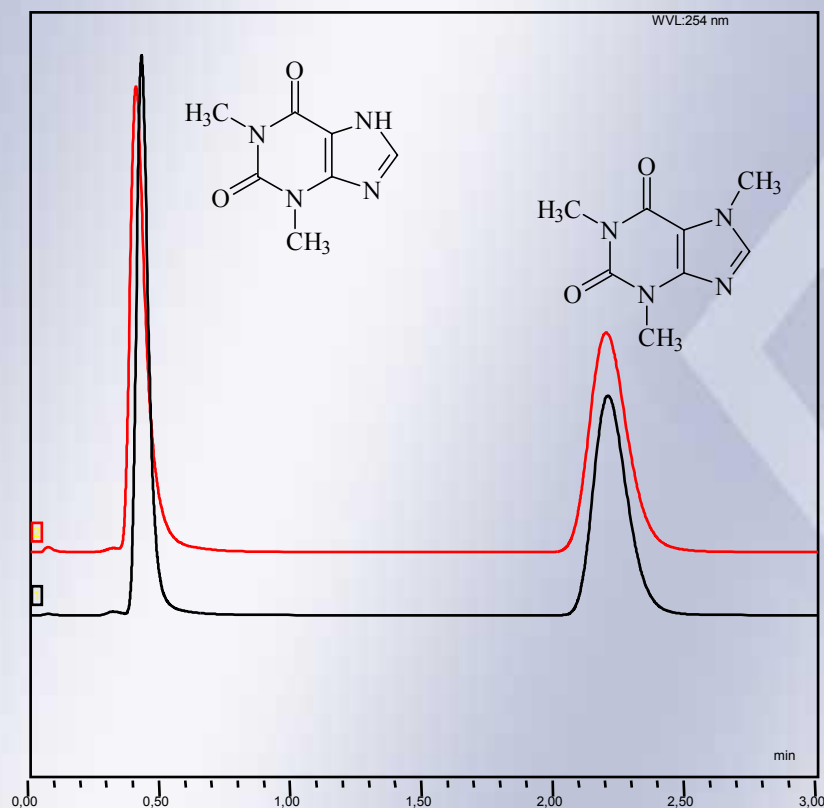
Flow: 1 ml/min

Temp.: 30 °C

Detection: UV, 230 nm

Sample: Pyridine
Toluene
Ethylbenzene

RP 18 phases under basic pH



Column: EC 50/4.6 Nucleodur **C18 Gravity**, 5 μm

Eluent: MeOH/H₂O/NH₃
20:80:0,5 (v/v/v), **pH 11**

Temp.: 30°C

Flow: 1.3 ml/min

Detection: UV, 254 nm

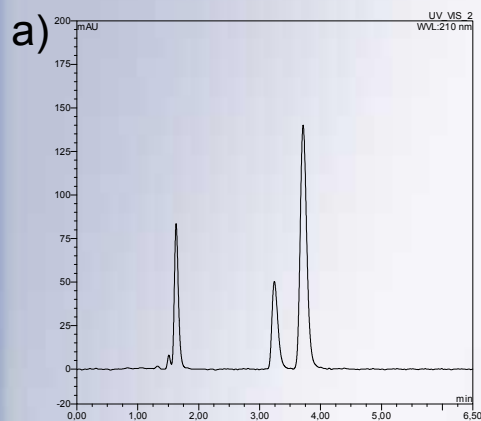
Sample: Theophylline
Caffeine

1. first injection
2. After 300 injections

Stationary phase selection

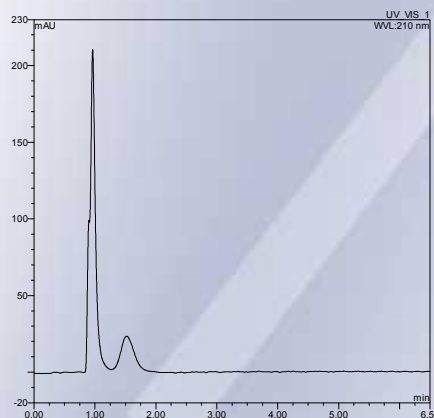
- Silica support
 - ✓ Mechanical properties/metal contaminants
- Surface chemistry
 - ✓ Base deactivation
 - ✓ Retention and selectivity
 - ✓ pH-stability and selectivity
 - ✓ **Stability in 100% aqueous eluents**
 - Special selectivity features

Stability test in 100% aqueous eluents

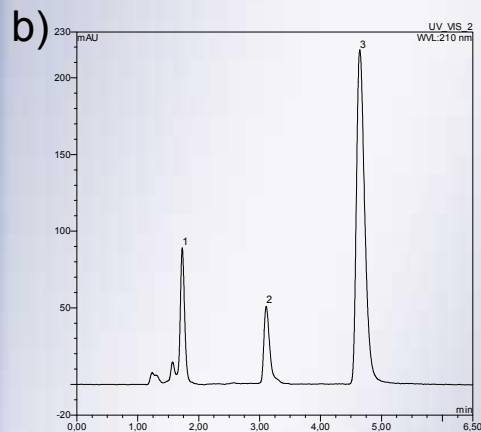


→ initial injection

pump
stopped!

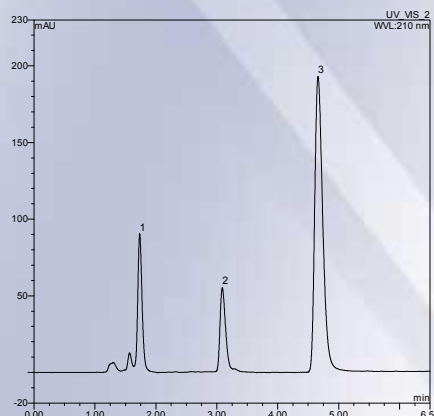


→ restart after 5 min



→ initial injection

pump
stopped!



→ restart after 12 h

Adsorbent: a) **Conventional RP 18 column**

b) **NUCLEODUR C18 Pyramid**

Column: 125 x 4 mm

Eluent: 50 mM KH_2PO_4 , pH 2.5

Flow rate: 0.7 ml/min

Temp.: 25 °C

Detection: UV, 210 nm

Inj. Vol.: 1 μl

1. Tartaric acid
2. Acetic acid
3. Maleic acid

- good stability in aqueous eluents
- similar selectivity features to RP18
- apparently lower bleeding characteristics (e.g. for MS detection)

Stationary phase selection

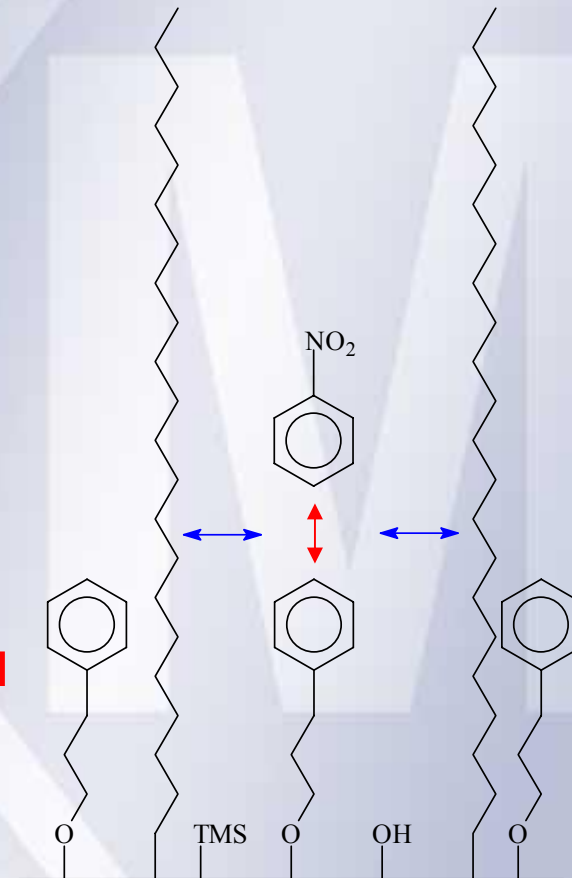
- Silica support
 - ✓ Mechanical properties/metal contaminants
- Surface chemistry
 - ✓ Base deactivation
 - ✓ Retention and selectivity
 - ✓ pH-stability and selectivity
 - ✓ Stability in 100% aqueous eluents
 - ✓ **Special selectivity features**

Phenyl Octadecyl phase

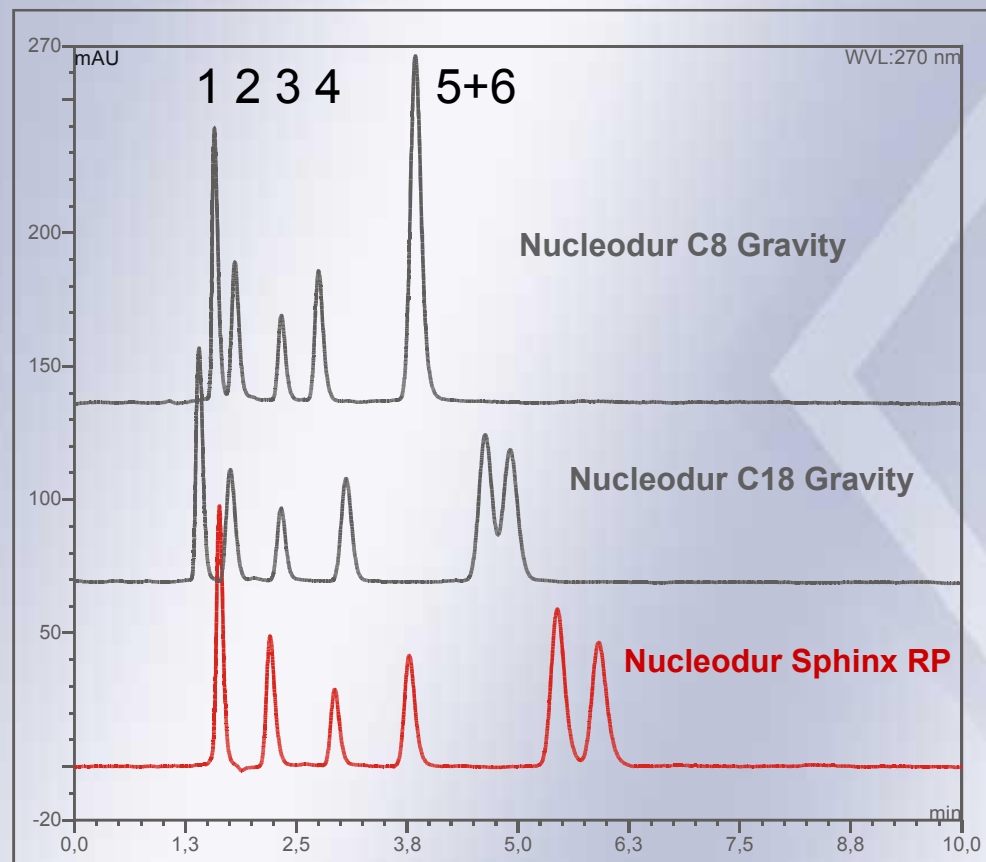
Unique Concept of
Nucleodur **Sphinx RP** !

Optimized proportion of octadecyl- and
phenylpropylsilanes

Hydrophobic and π - π interactions combined



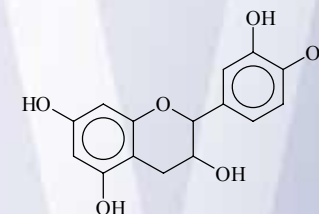
Separation of Flavonoids



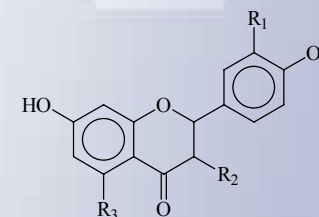
Chromatographic conditions:

150/4.6 mm Nucleodur **Sphinx RP**, 5 μ m
 methanol/water 40:60 (v/v)
 1 ml/min, UV, 270 nm, 30 °C

1. Catechin



2. Rutin $R_1 = R_3 = \text{OH}$, $R_2 = \text{O-Rutinose}$
3. Fisetin $R_1 = R_2 = \text{OH}$, $R_3 = \text{H}$
4. Quercetin $R_1 = R_2 = R_3 = \text{OH}$
5. Kaempferol $R_1 = \text{H}$, $R_2 = R_3 = \text{OH}$
6. Isorhamnetin $R_1 = \text{OMe}$, $R_2 = R_3 = \text{OH}$



C18 phase with steric selectivity

- Steric selectivity is the ability to separate compounds in relation to its molecular structure and geometry.
- Compounds with rigid structure show a steric behaviour:

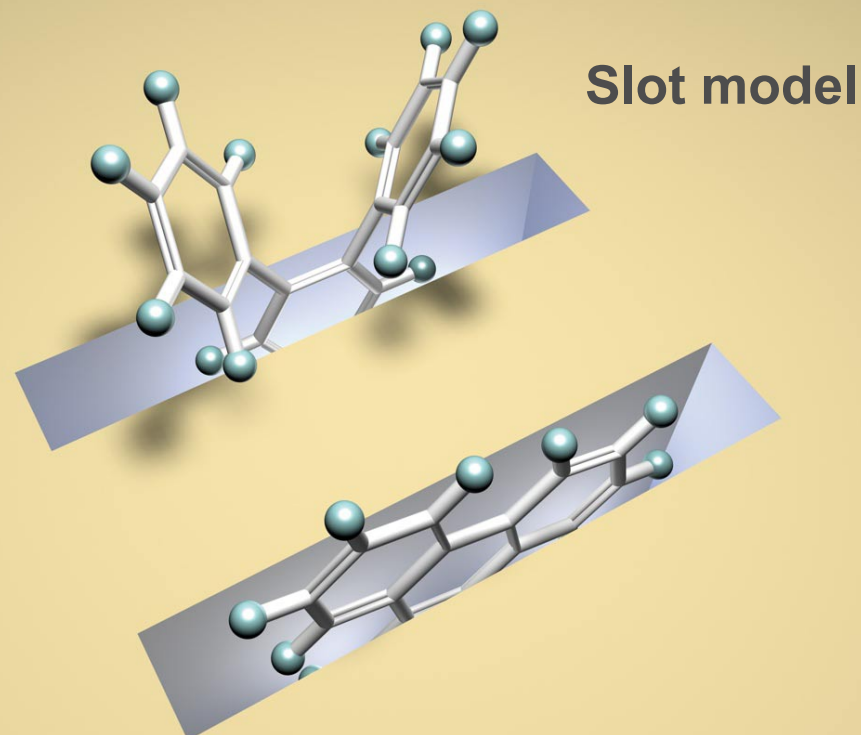
e.g.: polycyclic compounds (PCB)

steroids

carotinoides

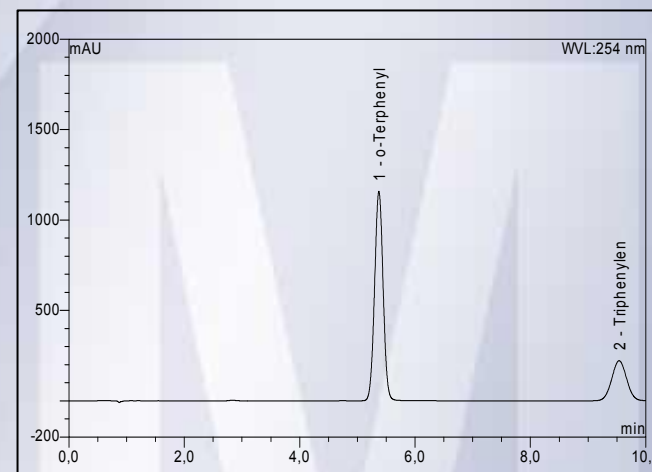
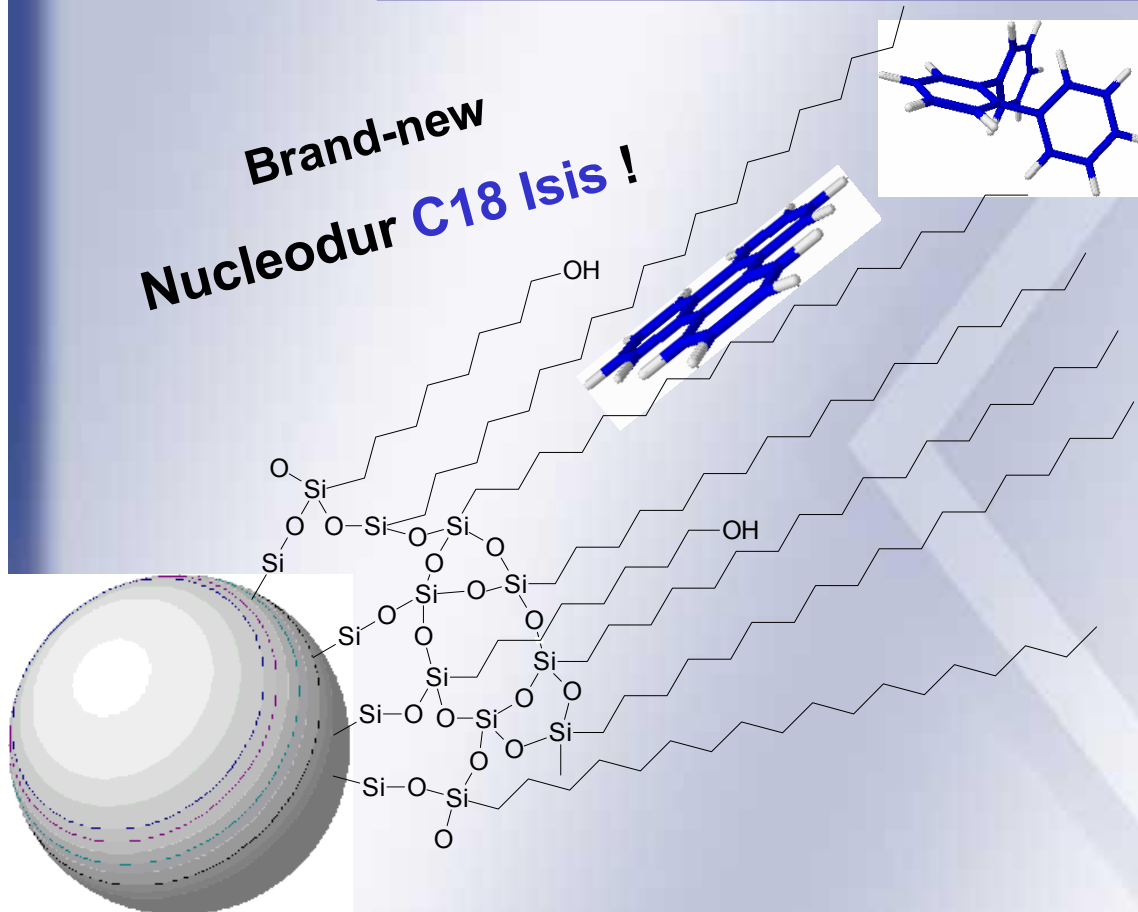
vitamines (D2/D3)

aliphatic compounds with rigid structure (double bonding)



C18 phase with steric selectivity

Brand-new
Nucleodur **C18 Isis** !



$\alpha_{\text{Triphenylene/o-Terphenyl}} = 1,93 !$

column: 125 x 4 NUCLEODUR **C18 Isis**

eluent: methanol / water (80:20)

flow rate: 1 ml/min

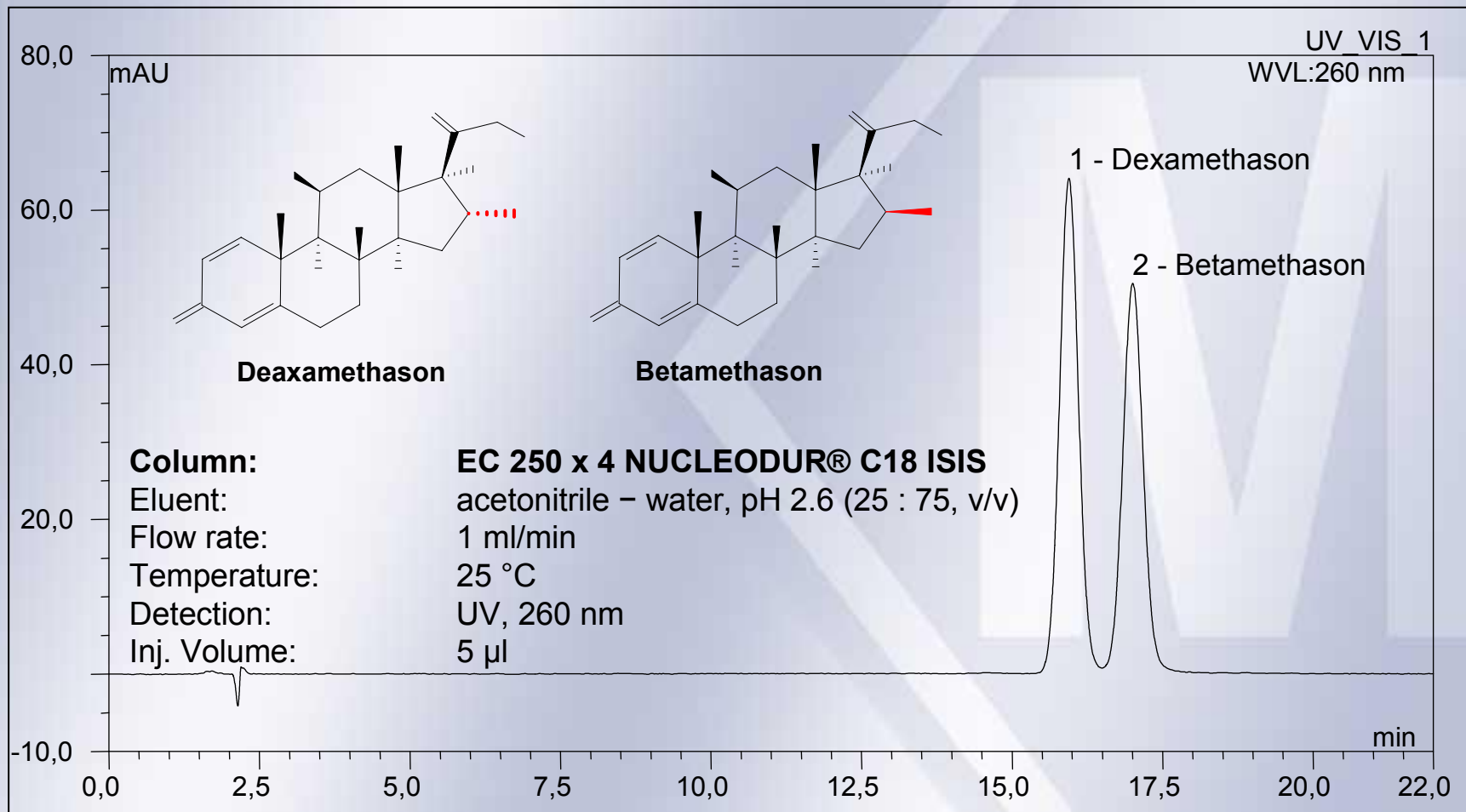
temp.: 40 °C

detection: UV, 254 nm

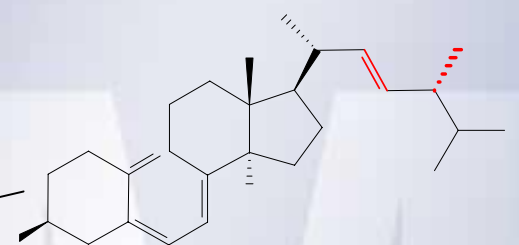
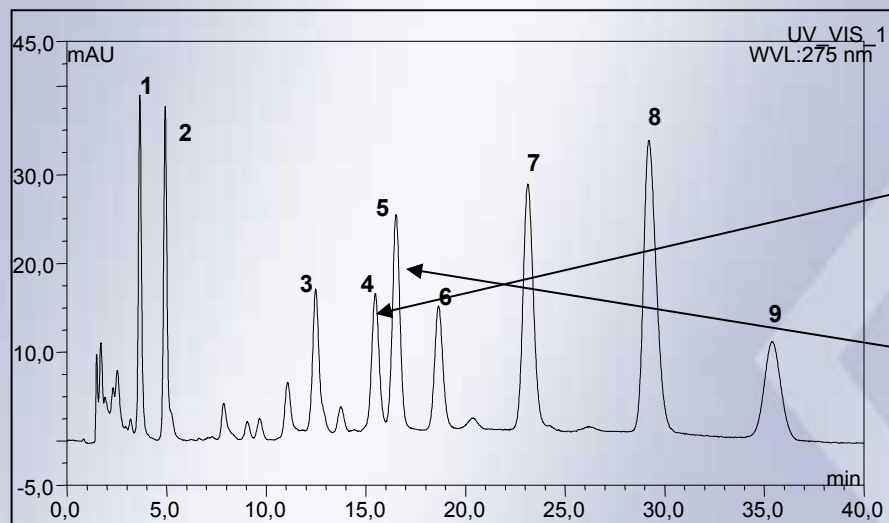
For standard RP 18 phases:

$\alpha_{\text{Triphenylene/o-Terphenyl}} = 1,0 - 1,2$

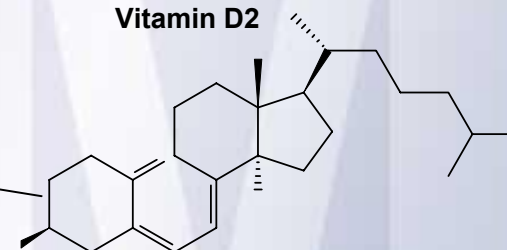
Dexamethason / Betamethason



Fat-soluble vitamins



Vitamin D2



Vitamin D3

Column: EC 125 x 2 mm NUCLEODUR® C18 ISIS
Eluent: acetonitrile / water (100:5, v/v)
Flow rate: 0.2 ml/min
Temperature: 25 °C
Detection: UV, 275 nm
Inj. Volume: 5 µl

- 1) Vitamin A
- 2) Vitamin A acetate
- 3) Vitamin K2
- 4) Vitamin D2
- 5) Vitamin D3
- 6) γ-Tocopherol
- 7) α-Tocopherol
- 8) α-Tocopherol acetate
- 9) Vitamin K1

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